

Monitoring of Seasonal Variations in Radiocesium from the Fukushima Accident in Litterfall of Four Broad-Leaf Tree Species and its Correlation with Variation in Potassium

Toshihiro Yoshihara*, Hideyuki Matsumura and Shin-Nosuke Hashida

Plant Molecular Biology, Laboratory of Environmental Science, Central Research Institute of Electric Power Industry (CRIEPI), Japan

Research Article

Received date: 24/05/2016

Accepted date: 14/10/2016

Published date: 18/10/2016

*For Correspondence

Toshihiro Yoshihara, Plant Molecular Biology, Laboratory of Environmental Science, Central Research Institute of Electric Power Industry (CRIEPI), 1646 Abiko, Chiba 270-1194, Japan, Tel: +81-4-7182-1181.

E-mail: yoshiha@criepi.denken.or.jp

Keywords: Radiocesium, Potassium, Fukushima, Tree, Translocation

ABSTRACT

The concentration of radiocesium (^{137}Cs) primarily derived from the Fukushima accident in March 2011 was measured in litterfall of four broad leaf trees (*Prunus x Yedoensis* cv. Somei-Yoshino, *Cinnamomum camphora*, *Acer buergerianum*, and *Aesculus hippocastanum*) to see whether a typical reduction would occur during defoliation as with its biologically essential analogue, potassium (K). The sampling was performed every two to four weeks between 2011 and 2014, using a trap to prevent contact with the ground. Results indicated that the concentrations of ^{137}Cs basically decreased over time in each defoliation season, although a small derangement could be seen. On the other hand, the concentrations of K in litterfall (except for Somei-Yoshino) once increased at the mid-season and then decreased. As a result, although the K concentrations in litterfall collected last were less than that collected earlier in the season, the seasonal changes in K and ^{137}Cs were not synchronized except for Somei-Yoshino. It is possible that an unknown species-specific K regulation-based time-lag and/or irregularity could influence the synchrony between ^{137}Cs and K transportation, although the seasonal reduction in ^{137}Cs concentration would typically occurred as part of the physiological demand for K.

INTRODUCTION

Trees can be divided into two classes according to their leaf longevity: deciduous and evergreen. Leaf longevity is an adaptive and/evolutionary strategy that enables species to survive in a range of different environments. When faced with drought or cold stress, for example, deciduous species shed their leaves to minimize the detrimental effects. In contrast, by retaining their leaves with much advanced tolerance against stresses, evergreen species expend less energy in developing new leaves. However, all trees eventually shed their leaves. This is a necessary and constructive step in the life cycle of trees, which is required for the renewal of senescent parts, optimization of the spatial arrangement of leaves, and as a competitive measure against neighboring individuals^[1]. Prior to shedding their leaves, trees typically translocate nutrients from senescent leaves to the tree body for recycling^[2,3]. It is known that translocation efficiency (=resorption efficiency; a specific elemental concentration in green foliar parts [A] minus that in litterfall [B] is divided by [A]) is dependent on both the type of element and tree species. In this regard, compared resorption efficiencies for nine essential elements (N, P, K, S, Mg, Ca, Cu, Fe, and Mn) among four tree species (Birch, Ash, Oak, and Lime) grown in the same soil and under the same climate conditions^[4]. Among the findings of this study were the following: the highest resorption efficiency in Birch was for N (more than 70%), whereas the lowest was for Ca and Mg (<5%); Lime trees have 10% to 20% higher resorption efficiency for P and K than the other species, while Ash trees have 15% to 40% lower efficiency for N and S than the other species. In addition to these basic differences according to element and tree species, certain other factors, such as soil fertility, leaf nutrient status, annual precipitation, time span of senescence, and summer temperature, have been reported to influence nutrient resorption^[5,6].

Non-essential elements also accumulate/recycle in trees at certain ratios. These non-essential elements can enter trees incidentally via the same uptake/translocation systems as essential elements^[7]. For example, Cd is easily translocated from the soil to any part of a tree via the various uptake pathways of Fe^{2+} , Zn^{2+} , and other essential cations. Thus, uptake/translocation of non-essential elements is probably dependent upon soil conditions, particularly in relation to the abundance of related essential elements, and tree species. In other words, whenever considering non-essential elements, it is necessary to know the status of related essential elements. In this regard, although to date the relationships between Cd and divalent metal ions have been

particularly well documented from ecological to molecular levels, there is still limited knowledge on the relationships of other non-essential elements^[7]. With respect to such incidental uptake/translocation, the status of harmful elements, such as Cd, Hg, As, Se, and Pb, is particularly important for our lives and forest ecosystems, because of their possible detrimental impacts and/or frequency of contamination^[8,9].

In addition to the influences of these conventional elements, there have been growing concerns about the fate of radionuclides in forests following the accident at the Fukushima Dai-ichi nuclear power plant in March 2011^[10]. Radiocesium (¹³⁴Cs and ¹³⁷Cs) contamination is of particular concern, because of the comparatively longer physical half-lives of these radionuclides (more than 2 and 30 y, respectively) and their abundance in the Fukushima fallout^[11]. Furthermore, forests are known to be effective interceptors of radionuclides compared with other land-use areas^[12,13]. However, with the exception of those in the vicinities of residential areas, forests have unfortunately been excluded from the governmental decontamination plan for economic reasons^[10]. Because Cs is a non-essential element, when assessing the fate of radiocesium in forests, it is necessary to take into consideration its relationship with essential elements. As a biological analogue of Cs, the metabolism of K in trees is most closely related to that of Cs^[14]. Accordingly, sufficient K fertilization can decrease radiocesium accumulation in trees, whereas K deficiency may increase accumulation^[15]. In forest ecosystems, forest soils basically tend to be K deficient, although the K level varies widely with seasons and/or individuals^[16]. This may emphasize the importance of K recycling, which should affect the status of radiocesium. Indeed, we have demonstrated a strong correlation between radiocesium and K and a possible synchronized resorption in Japanese flowering cherry during leaf senescence^[17]. In contrast, litterfall of Japanese cedar showed little correlation between radiocesium and K (Yoshihara et al, although a strict correlation could be seen in age-classified green leaves^[18]). This may reflect the prolonged leaf longevity of Japanese cedar and the related physiology, and thus, that the fate of radiocesium in the environment is highly dependent on the leaf ecology of each tree species. Further understanding of this relationship (e.g., in terms of species specificity and effect of environmental conditions) would not only be important for demonstrating accumulation and cycling of a specific element in a forest ecosystem from an ecological perspective, but would also be indispensable for assessing the potential impact on human health, and for establishing countermeasures. This paper demonstrates the results from one such trial to elucidate species specific correlation between K and radiocesium in some tree species (i.e., Flowering cherry, Camphor tree, Horse chestnut, and Trident maple) during their leaf senescence. These trees are popular broad leaf woody species in Japan. Horse chestnut, Flowering cherry, and trident maple are deciduous but Camphor is a kind of ever green. A comparative study among these, and previous results from flowering cherry and Japanese cedar should provide us a good recognition of the fate of radiocesium in trees^[18,19].

MATERIAL AND METHODS

Tree Species and Sampling

The sampling targets were four broad leaf trees, some of the most popular ornamental woody species in Japan (*Prunus x yedoensis* cv. Somei-Yoshino (Flowering cherry), *Cinnamomum camphora* (Camphor), *Aesculus hippocastanum* (Horse chestnut), and *Acer buergerianum* [Trident maple]). These trees stood alone or as a part of colonnades and grew on bare ground at our laboratory in Abiko, Chiba (Laboratory of Environmental Science, CRIEPI;^[17,19,20] **(Figure 1)**. Abiko has a moderate monsoon climate (the average of 2011 and 2014 total annual rainfall and annual daily temperature were 1419.8 mm and 14.4 °C, respectively; Japan meteorological agency, <http://www.jma.go.jp/jma/menu/report.html>). The basal soil type was pale Ando soil overlaid with a thin organic layer^[21]. Litterfall, consisting primarily of fallen leaves, was sampled between 5 September 2011 and 28 April 2014 **(Figure 2)**. It is noteworthy that some data of Flowering cherry (i.e., 2011 and 2012) was obtained from that presented in^[19]. Basically, four to eight samplings were conducted during each defoliation season (at two-week intervals), although the interval was sometimes extended up to three months, depending on the sample volume. Independent samples from at least two to five stands were collected and independently measured with respect to the concentrations of radiocesium and potassium, as shown below. Two box-type traps (each 1 × 1 × 1 m³) were randomly set for each stand to collect litterfall before the litter could contact the ground, and litterfall collected with two traps at the same time from the same stand were mixed together and treated as one sample **(Table 1)**. Every sample was dried for more than 3 days in an oven at 65 °C, weighed, and then kept in the oven until further analyses were performed.

Analysis

Each sample was weighed, homogenized, and packed in containers (U-8 container, 56 mm diameter × 68 mm height, As-one Co. Ltd., Osaka, Japan) for radioactivity measurement^[22]. A Ge-semiconductor detector (GEM20-70, ORTEC, Oak Ridge, TN: The detection limit for 50000 s is about 0.001 Bq g-DW⁻¹), coupled with a multichannel analyzer (MCA7600, SEIKO EG&G, Tokyo, Japan), was used to evaluate ¹³⁷Cs activity in the samples. For samples in which radiocesium was undetectable with the 3600s measurement, another 30000s measurement was conducted. Radiocesium activities were rounded off to ten digit or less following the correction of physical decay to the date of assumed fallout^[23]. After the nuclide analysis, a 0.5 g aliquot of each dried sample was wet-ashed with a solution of HNO₃ and H₂O₂ for two nights at 110 °C and dissolved in 15 ml of 1 N HCl. Elemental concentrations were measured using inductively coupled plasma atomic emission spectrometry (ICP-OES; Optima 5300DV, Perkin-Elmer, Waltham, MA) at wavelengths of 766.490 (K) nm. To compare the differences among each mass of

data, two factors ANOVA, Dunnett’s multiple comparison test, and linear regression analysis was performed with an add-on software package for Microsoft Excel, Excel-statistics (ver. 7.0; Social Survey Research Information Co., Ltd., Tokyo, Japan) and/or Origin (ver. 8.1J SR3; OriginLab Co., MA, USA). In addition, tests of equality between sets of slopes and intercepts in two linear regressions were performed to evaluate the significance of differences in correlations between ¹³⁷Cs and K concentrations [24].

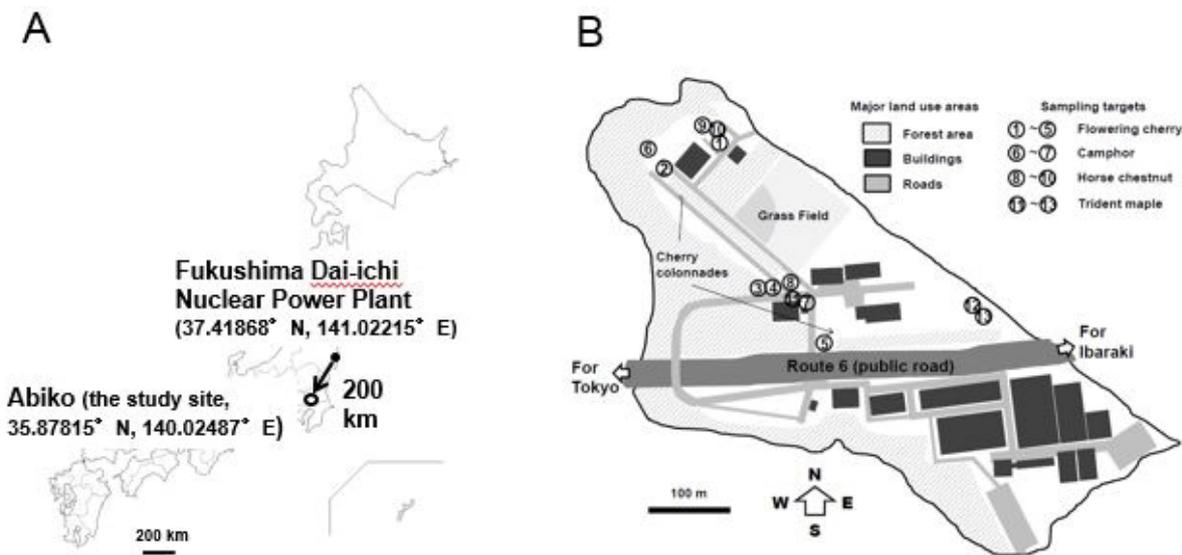


Figure 1. Schematic description of the experimental site A. Location of the study site relative to the Fukushima Daiichi nuclear power plant; B. Distributions of the target stands within the study site.

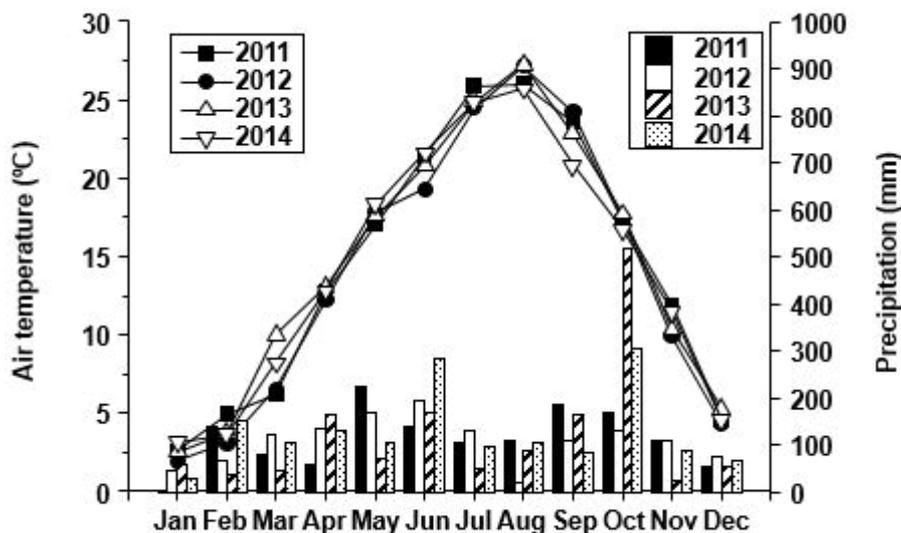


Figure 2. Meteorological data in Abiko between 2011 and 2014. Annual changes in the monthly average of air temperatures and monthly integrated 455 precipitation in Abiko are indicated for the observation period of 2011 to 2014.

Table 1. A list of the observed plant species.

Observed species	Number of observed stands (A) ^{*1} (litter traps)	Total sampling times per defoliation season (B) ^{*2}				Total samples per species Σ (A) × (B) ^{*3}
		2011	2012	2013	2014	
1. Flowering cherry (<i>Prunus x yedoensis</i> cv. Somei-yoshino)	5 (10)	5	6	5	-	80
2. Camphor tree (<i>Cinnamomum camphora</i>)	2 (4)	-	6	8	-	28
3. Horse chestnut (<i>Aesculus hippocastanum</i>)	3 (6)	-	4	6	-	30
4. Trident maple (<i>Acer buergerianum</i>)	3 (6)	-	6	6	-	36

1. Number of observed stands (A) are basically equal to the number of replication (n) for each averaged data in the following figures. Two box-type traps were set for each stand to collect litterfall, and litterfall collected with two traps at the same time from the same stand were mixed together and treated as one sample. 2. Total sampling times per defoliation season (B) were shown by each stand. It is noteworthy that the defoliation season of Camphor is extending over year, thus the numbers are indicated between 2012 and 2013 or 2013 and 2014, respectively. 3. Total samples per species were shown by each of (A) Q(B).

RESULTS AND DISCUSSION

Autumnal Changes in Radiocesium Activity in Litterfall

The result of ANOVA indicated that the effect of a year-to-year variation (i.e., 2011, 2012/2012–13 and 2013/2013-2014 defoliation seasons) was significant ($p \leq 0.001$; **Table 2**) on ^{137}Cs concentrations in litterfall, although the effect of plant species and the combination of these factors were not significant ($p \geq 0.05$; **Table 2**). Although these results were based on litterfall samples, it is interesting that these corresponded to the previous report for ^{137}Cs concentrations in current year green leaves (i.e., leaves still living on tree) between 2012 and 2013 in some broad-leaf tree species [17]. On the other hand, it is commonly noted in the Fukushima accident studies that the radiocesium concentration in evergreen species is higher than in Deciduous species and the annual rate of decrease during the early phase of fallout (i.e., usually around two years) is larger in evergreen species than in deciduous tree species [20,25-28]. These are generally explained by a difference in the ecological characteristics of trees; particularly, whether their foliage was expanded at the time of the fallout and directly received radiocesium deposition. In this regards, Camphor tree has broad leaves but is a kind of evergreen species, and ordinarily grows leaves in cycles of more than one year [29]. Thus, unlike other evergreen species, the litterfall of Camphor tree seemed to be already outside of the interval of influence from direct deposition, as shown in ordinary deciduous species at least in the 2012-2013 defoliation season.

Table 2. Two factors ANOVA to determine the effects of plant species and defoliation seasons on ^{137}Cs and K concentrations in litterfalls.

	Plant species (A)		Defoliation season (B)		(A) × (B)	
	F	P	F	P	F	P
^{137}Cs concentration	0.16	0.92	19.3	≤ 0.001	1.13	0.34
K concentration	11.8	≤ 0.001	3.00	0.05	1.80	0.15

Concentration data for individual samples between 2011 and 2013 defoliation seasons (in total of 174) were used for the calculations. Results indicate that sampling season exerted a highly significant effect on K concentration and plant species also exerted a highly significant effect on K concentration, although the combinatorial effect was not significant in both the concentrations

The ^{137}Cs concentration in litterfall basically exhibited a time-dependent decrease in every observed species (**Figure 3**). In fact, the reductions in the concentration from the beginning to the end of each defoliation season were 250-530 Bq kg-DW-1, 190-650 Bq kg-DW-1, 350–1270 Bq kg-DW-1, and 90-680 Bq kg-DW-1 in Flowering cherry, Camphor tree, Horse chestnut, and Trident maple, respectively. In particular, the reduction in Flowering cherry was significant in every observed season from 2011 to 2013 (**Figure 3A**). Thus, the previous reports for an autumnal decrease of ^{137}Cs concentration in cherry litterfall was clearly ascertained in both the following seasons and expanded numbers of species [17]. It is noteworthy that the duration of decrease was prolonged and the timing was deviated in evergreen Camphor trees than those of the other observed species, although the ^{137}Cs concentrations decreased at the end of each defoliation season (**Figure 3B**). Such disparities may reflect the leaf longevity and defoliation timing of Camphor tree. However, in case of another evergreen species, Japanese cedar showed no specific seasonal decrease in the ^{137}Cs concentration in litterfall [18]. It is possible that extended leaf longevity of the other evergreen species than Camphor tree related to this non-specific seasonal decrease in litterfall (**Figure 3C**). Otherwise, this might be related to a morphological and/or physiological difference between conical and broad-leaf species. On the other hand, the possibility cannot be ruled that a late starting time for the sampling (i.e., late September) in 2012 could have induced apparent insignificance for the decrease in every species except Flowering cherry. In addition, a few irregular changes in the ^{137}Cs concentrations were observed in the middle of the defoliation season, particularly in Trident maple (**Figure 3D**). Some factors independent of tree physiology, such as accidental incorporation of higher radiocesium containing tissues (e.g., bark) in samples, or strict fixation of radiocesium in specific tissues, could relate to the irregular changes. Otherwise, it is possible that a temporal derangement in physiological orders related to this phenomenon. Nevertheless, it is probable that the seasonal decrease of ^{137}Cs concentration in litterfall would be a common phenomenon, at least among observed tree species. If the seasonal decrease of ^{137}Cs concentration in litterfall could be considered a physiological matter, the mechanism should be shared with other elements, which are essential for plant growth, and which behave as biological analogues of Cs. As one such possible mechanism of the seasonal change in ^{137}Cs concentration, the changes in K concentrations is discussed in the following section.

Autumnal Changes in Potassium Concentration in Litterfall

The result of ANOVA indicated that the effect of plant species was significant ($p \leq 0.001$; **Table 2**) on K concentrations in litterfall, but the effect of a year-to-year variation and the combination of these factors were not significant ($p \geq 0.05$; **Table 2**). A species specific and year-to-year variation in K concentration of foliar parts is a commonly observed phenomenon [16,30,31]. On the other hand, Flowering cherry demonstrated a clear time-dependent decrease in the K concentration in every season from 2011 to 2013 (**Figure 4**). Such a seasonal decrease of K concentration in litterfall is also well-known [2,3,32-34]. A decrease of K in living leaves is paralleled by that in freshly fallen leaf litter and time-dependent increases in nutrient resorption efficiency [35] (Niinemets a). Although two factors in tree physiology, resorption and leaching, could be a major cause of the nutrient decrease, recent observations concluded that resorption accounts for larger contribution than does leaching [4,36,37]. The resorption efficiency varies by element, plant species, and nutritional/soil conditions. The resorption ratio of K is usually between 30% and 90% [4,34,35,38,39].

In contrast to the typical time-dependent decrease shown in Flowering cherry, the seasonal decrease in litterfall recognized

in Camphor tree, Horse chestnut, and Trident maple was distinct (**Figure 3B-3D**). That is, in these species, the K concentrations showed the highest values at the mid-point of each defoliation season (i.e., “a mid-season increase”), although the lowest values were shown at the end (except for Camphor tree in 2012-2013). The amount of reductions in the concentration from the beginning to the end of each defoliation season were -0.96 to 0.32 mg g-DW⁻¹, 0.59-2.24 mg g-DW⁻¹, and 0.75-1.99 mg g-DW⁻¹ in Camphor tree, Horse chestnut, and Trident maple, respectively (**Figures 3B-3D**). These reductions were considerably less than in Flowering cherry (3.21 to 9.07 mg g-DW⁻¹; **Figure 4**). Such a “mid-season increase” in the K concentration in litterfall can be seen in some other species, such as pine, oak, hickory, and mesophytic hard woods [33]. In this regard, observed a time-dependent decrease in K content of freshly fallen leaf litter in most, but not in all species [35]. They hypothesized that time dependence could arise if the leaves that had been shed at different times, had also entered senescence at different times, but had undergone resorption simultaneously, might result in equal nutrient withdrawal from the leaves. Conversely, leaves that had not kept the pace of resorption that others had, would not show the time-dependent decrease. In addition to such lack of coordination in related physiological processes (i.e., shedding, senescence, and resorption), another hypothesis, such as reflux and/or overflow of once resorbed elements from the sink, cannot be eliminated from attempts to explain the “mid-season increase”.

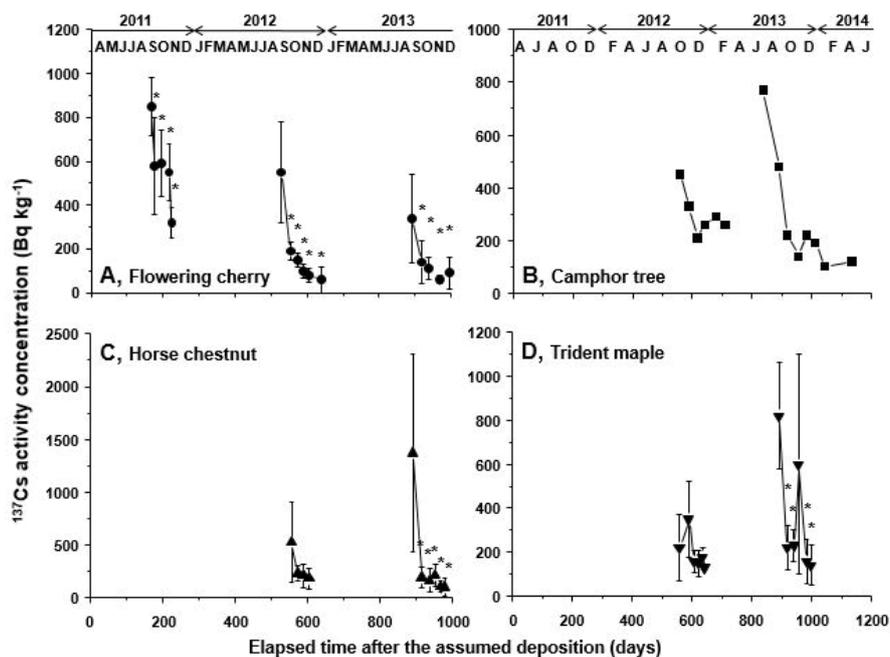


Figure 3. Transitions of radiocesium concentration in litterfall of four broad-leaf tree species. The means and standard deviations of ¹³⁷Cs activity concentration (kBq kg-DW⁻¹) in individual samples (n=2 to 5, **Table 1**) are shown by sampling season: A, Flowering Cherry; B, Camphor tree; C, Horse Chestnut; and D, Trident Maple. The ¹³⁷Cs activity concentration is displayed after adjustment for the date of assumed deposition (21 March 2011). The horizontal axis is indicating the elapsed time after the assumed deposition (days) with year and initial of corresponding month in upper space. Asterisks on the standard deviations indicate significant differences ($p \leq 0.05$) between data of the first sampling in each defoliation season and later ones, using Dunnett’s multiple comparison tests. Data for Camphor tree are not provided for the statistical analysis because replication was insufficient. The data from 2011 and 2012 for flowering cherry was obtained from that presented in Yoshihara et al.

Synchrony in Seasonal Changes Between Radiocesium and Potassium Concentration in Litterfall

To see a specific correlation between the ¹³⁷Cs and K concentrations, individual sets of pairwise data before averaging were provided for linear regression analysis. Results indicated that a significant correlation between ¹³⁷Cs and K concentration was demonstrated in Flowering cherry in every year observed (**Figure 5 and Table 3**). The Pearson’s r of each regression line in 2011, 2012, and 2013 were 0.45 ($p=0.025$), 0.83 ($p \leq 0.001$), and 0.56 ($p \leq 0.01$), respectively. Thus, the time dependent decrease in the ¹³⁷Cs concentration was highly correlated with that of the K concentration in Flowering cherry (i.e., high synchrony), and this correlation was correspondent to that in previous reports [19]. It is very interesting that the similarity in regression slopes also remained through these years (i.e., 49, 49, and 44 in 2011, 2012, and 2013, respectively; $p \geq 0.05$). However, the regression intercepts were similar between 2012 and 2013 ($p \geq 0.05$), but different between 2011 and the others ($p \leq 0.05$). This could be due to reduction of the ¹³⁷Cs concentration within 2011 by an environmental cause, probably the so-called weathering effect. However, this effect disappeared after 2012; so not much difference was seen in the ¹³⁷Cs concentrations in 2012 and 2013.

In contrast to the case of Flowering cherry, no significant correlation between ¹³⁷Cs and K concentration could be observed in Camphor tree, Horse chestnut, and Trident maple (**Figure 5 and Table 3**). The Pearson’s r of each regression line was up to 0.51 ($p=0.093$) in the 2012-2013 season for camphor tree, and some values were almost zero (e.g., $r=0.002$, $p=0.99$; in 2013 in Horse chestnut) or minus (e.g., $r=-0.19$, $p=0.60$; in 2012 in Trident maple). Similar values in the regression intercept were obtained for each species (**Table 3**). Similarly non-significant correlation between ¹³⁷Cs and K concentration could also be seen in preceding observations. For example, Antonopoulos-Domis et al. Demonstrated that 40K concentrations in green leaves of apricot trees showed time-dependent decrease during the growing seasons [40,41], whereas the ¹³⁷Cs concentration was maintained at its initial

value throughout the season [42-44]. However, in the present case, a time-dependent decrease in the ^{137}Cs concentration was more or less indicated in every observed species, but a decrease in the K concentration was differently indicated after the “mid-season increase”. In addition, although the time of appearance was different, a similar “mid-season increase” could be seen not only in the K concentration but also in the ^{137}Cs concentration in Trident maple. Together all, it is hypothesized that a basic mechanism to resorb ^{137}Cs and K was at work, but that an unknown factor or factors with particular effect on K, could have influenced the correlation and disrupted the synchrony. As mentioned in the previous section, the results could be due to lack of coordination in related physiological processes (i.e., shedding, senescence, and resorption), or due to participation of another mechanism, such as reflux and/or overflow of once resorbed elements from the sink.

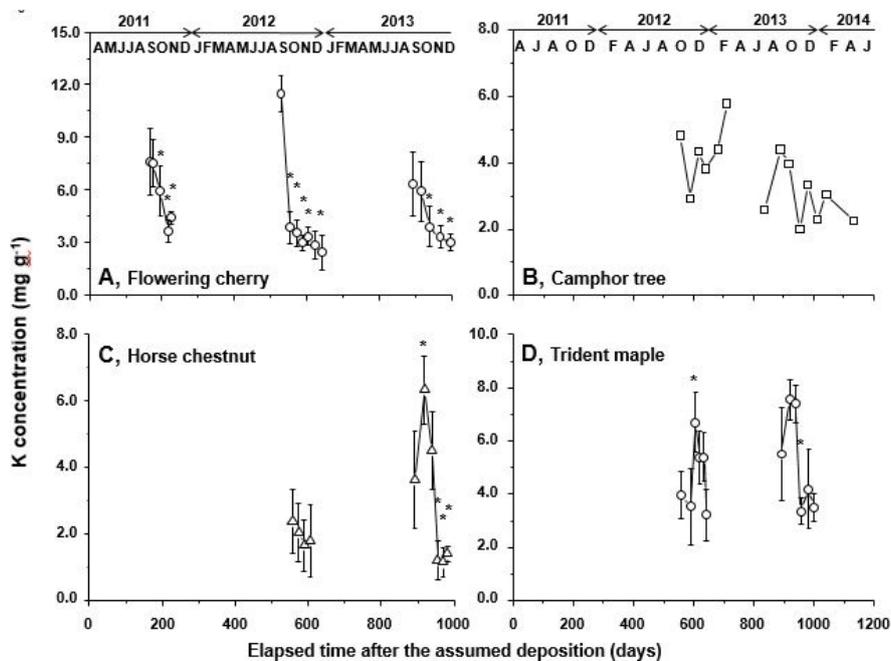


Figure 4. Transitions of potassium concentration in litterfall of four broad-leaf tree species. The means and standard deviations of K concentration (mg g-DW^{-1}) in individual sampling ($n=2$ to 5 , **Table 1**) are shown by sampling season: A, Flowering Cherry; B, Camphor tree; C, Horse Chestnut; and D, Trident Maple. Asterisks on the standard deviations indicate significant differences ($p \leq 0.05$) between data of the first sampling 475 in each defoliation season and later ones, using Dunnett’s multiple comparison tests. Data for Camphor tree are not provided for the statistical analysis because replication was insufficient. The data from 2011 and 2012 for Flowering cherry was obtained from that presented in Yoshihara et al.

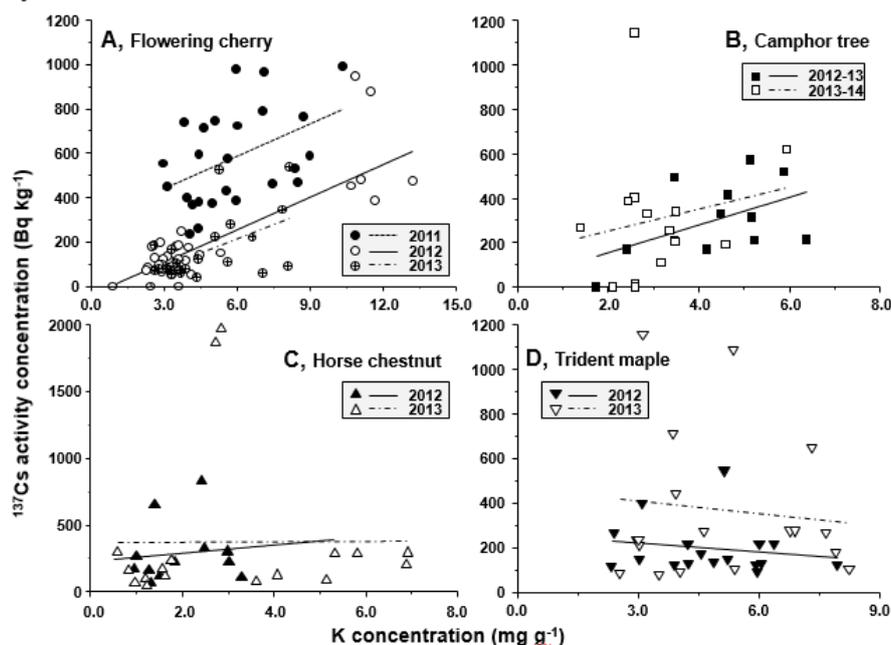


Figure 5. Correlations between ^{137}Cs and K concentrations 479 in litterfalls of four broad-leaf tree species individual sets of pairwise data for ^{137}Cs and K concentrations before averaging are indicated by each sampling season: A, Flowering Cherry; B, Camphor tree; C, Horse Chestnut; and D, Trident Maple. Results of linear regression analysis are demonstrated with dotted line (2011 season), solid line (2012 or 2012-2013 season), and one dot chain line (2013 or 2013-2014 season), respectively. The numerical expressions and the results of statistical analyses are presented in **Table 3**. The data from 2011 and 2012 for Flowering cherry was obtained from that presented in Yoshihara et al.

Table 3. Linear regression analysis of the correlations between ¹³⁷Cs and K concentrations in litterfalls.

Observed species	Defoliation season	Number of provided samples ^{*1}	Linear regression analysis ^{*2}				
			Intercept	Slope	Pearson's	F	P
1. Flowering cherry (<i>Prunus x yedoensis</i> cv. <i>Somei-yoshino</i>)	2011 ^{*3}	25	296	49	0.45	5.73	0.025
	2012 ^{*3}	30	37	49	0.83	61.71	≤0.001
	2013	25	48	44	0.56	10.07	≤0.01
2. Camphor tree (<i>Cinnamomum camphora</i>)	2012-13	12	28	63	0.51	3.44	0.093
	2013-14	16	152	49	0.18	0.42	0.53
3. Horse chestnut (<i>Aesculus hippocastanum</i>)	2012	12	230	29	0.10	0.11	0.75
	2013	18	371	0.54	0.002	6.04E-5	0.99
4. Trident maple (<i>Acer buergerianum</i>)	2012	18	264	-14	-0.19	0.60	0.45
	2013	18	466	-19	-0.11	0.18	0.67

¹. Individual sets of pairwise data for ¹³⁷Cs and K concentrations before averaging, which shown in **Figure 4**, are provided for each linear regression analysis in each defoliation season. ^{*2}. Results indicate that only Japanese flowering cherry exerted significant positive higher correlations between ¹³⁷Cs and K concentrations. ^{*3}. The data of 2011 and 2012 for Flowering cherry was obtained from that presented in Yoshihara et al.

In conclusion, a time-dependent seasonal decrease was observed both in the concentrations of ¹³⁷Cs and K, and sometimes, a specific correlation could be seen between these concentrations; however, this phenomenon would be dependent on species and highly variable by a specific condition. The mechanism inducing such variability was still unclear. Further evidence and analysis will be necessary to understand the phenomenon^[45-47].

ACKNOWLEDGEMENTS

We thank Mr. Hiroshi Shimura, Ms. Mari Sato, and Ms. Miki Ueda (Co. Ltd. Ceres), Mr. Yukihiro Taemi, and Mr. Keita Yamaguchi (Co. Ltd. Denryoku Tech. Sys.) for their skillful assistance in sampling and analysis of radionuclides. This research was supported in part by a Grant-in-aid for Scientific Research from the Japan Society for the Promotion of Science (JSPS, 15H04621).

REFERENCES

- Kikuzawa K. Ecology of leaf longevity from individual leaves to ecosystems. Kyoritsu publication co. ltd., Tokyo Japan, 2005;212.
- Leopold AC. Senescence in plant development. Science. 1961;134:1727-1732.
- Chapin FS. The Mineral Nutrition of Wild Plants. Ann Rev Ecol Systemat. 1980;11:233-260.
- Hagen Thorn A, et al. Autumn nutrient resorption and losses in four deciduous forest tree species. Forest Ecol Manage. 2006;228:33-39.
- Del Arco JM, et al. Effects of site characteristics on nitrogen retranslocation from senescing leaves. Ecology. 1991;72:701-708.
- Cote B, et al. Increasing N and P resorption efficiency and proficiency in northern deciduous hardwoods with decreasing foliar N and P concentrations. Ann For Sci. 2002;59:275-281.
- Clemens S. Toxic metal accumulation, responses to exposure and mechanisms of tolerance in plants. Biochimie. 2006;88:1707-1719.
- Watmough SA and Dickinson NM. Dispersal and mobility of heavy metals in relation to tree survival in an aerially contaminated woodland soil. Environ Pollut. 1995;90:135-142.
- Vega FA, et al. Influence of soil properties on the sorption and retention of cadmium, copper and lead, separately and together, by 20 soil horizons: comparison of linear regression and tree regression analyses. J Hazard Mater. 2010;174:522-533.
- MOE (Ministry of the environment) 2012. Decontamination plans for forests, <http://josen.env.go.jp/about/efforts/forest.html>, In Japanese, Final access 2016/3/17.
- Morino Y, et al. Atmospheric behaviour, deposition, and budget of radioactive materials from the Fukushima Daiichi nuclear power plant in March 2011. Geophys Res Lett. 2011;38:1-7
- Bunzl K and Kracke W. Cumulative deposition of ¹³⁷Cs, ²³⁸U, ²³⁹U, ²⁴⁰Pu, and ²⁴¹Am from global fallout in soils from forests, grassland and arable land in Bavaria (FRG). J Environ Radioact. 1988;8:1-14.
- Petroff A, et al. Aerosol dry deposition on vegetative canopies. Part I: Review of present knowledge. Atmospheric Environ. 2008;42:3625-3653.
- Myttenaere C, et al. Modelling of Cs-137 cycling in forests: developments and research needed. Sci Total Environ. 1993;136:77-91.
- Kruyts N and Delvaux B. Soil organic horizons as a major source for radiocesium biorecycling in forest ecosystems. J Environ Radioact. 2002;58:175-190.

16. Tripler CE, et al. Patterns in potassium dynamics in forest ecosystems. *Ecology Letters*. 2006;9:451-466.
17. Yoshihara T, et al. Changes in radiocesium contamination from Fukushima in foliar parts of 10 common tree species in Japan between 2011 and 2013. *J Environ Radioact*. 2014a;138:220-226.
18. Yoshihara T, et al. Radiocesium contamination in living and dead foliar parts of Japanese cedar during 2011-2015. *J Env Radioact*. 2016;164:291-299.
19. Yoshihara T, et al. A time dependent behavior of radiocesium from the Fukushima fallout in litterfalls of Japanese flowering cherry tree. *J Environ Radioact*. 2014b;127:34-39.
20. Yoshihara T, et al. Radiocesium contaminations of 20 wood species and the corresponding gamma-ray dose rates around the canopies at 5 months after the Fukushima nuclear power plant accident. *J. Environ. Radioact*. 2013;115:60-68.
21. National Land Agency, Japan. 1983. Soil Map, Ryugasaki. <http://dbx.cr.chiba-u.jp/gdes/LUS/12chiba.html>
22. MEXT (Ministry of Education, Culture, Sport, Science, and technology). The manual for analysis of radiocesium, In; Issue No. 3 of a series of manuals for measurement of radiocesium concentration. Japan chemical analysis center, Chiba, Japan. 1976;21-27.
23. Doi T, et al. Anthropogenic radionuclides in the atmosphere observed at Tsukuba: characteristics of the radionuclides derived from Fukushima. *J Environ Radioact*. 2013;122:55-62.
24. Chow GC. Tests of equality between sets of coefficients in two linear regressions. *Econometrica: J Eco Soc*.1960;28:591-605.
25. Amano H, et al. Radiation measurements in the Chiba Metropolitan Area and radiological aspects of fallout from the Fukushima Dai-ichi Nuclear Power Plants accident. *J Environ Radioact*. 2012;111:42-52.
26. Tagami K, et al. Translocation of radiocesium from stems and leaves of plants and the effect on Radiocesium activities in newly expanded plant tissues. *J Environ Radioact*. 2012;111:65-69.
27. Hashimoto S, et al. The total amounts of radiocesium concentration contaminated materials in forests in Fukushima, Japan. *Scientific Rep*. 2012;2:2564
28. Koizumi A, et al. ¹³⁷Cs Trapped by Biomass within 20 km of the Fukushima Daiichi Nuclear Power Plant. *Environ Sci Technol*. 2013;47:9612-9618.
29. Cracker LE and Simon JE. Herbs, spices, and medicinal plants: recent advances in botany, horticulture, and pharmacology. Psychology Press, 1986;1:1-364.
30. Vitousek P. Nutrient cycling and nutrient use efficiency. *Am Nat*. 1982;119:553-572.
31. Killingbeck KT, et al. Foliar senescence in an aspen (*Populus tremuloides*) clone: the response of element resorption to inter-ramet variation and timing of abscission. *Can. J For Res*. 1990;20:1156-1164.
32. Guha MM and Mitchell RL. The trace and major element composition of the leaves of some Deciduous trees II. Seasonal changes. *Plant Soil*, 1966;24:90-112.
33. Grizzard TH, et al. Seasonal nutrient dynamics of foliage and litterfall on Walker Branch Watershed, a deciduous forest ecosystem. ORNL Oak Ridge National Laboratory (US) (1976).
34. Duchesne L, et al. Seasonal nutrient transfers by foliar resorption, leaching, and litterfall in a northern hardwood forest at Lake Clair watershed, Quebec, Canada *Can. J For Res*. 2001;31:333-344.
35. Niinemets U and Tamm U. Species differences in timing of leaf fall and foliage chemistry modify nutrient resorption efficiency in deciduous temperate forest stands. *Tree physiology*, 2005;25:1001-1014.
36. Wang WQ, et al. Seasonal changes in element contents in mangrove element retranslocation during leaf senescence. *Plant and soil*, 2003;252:187-193.
37. Milla R, et al. Relationships between phenology and the remobilization of nitrogen, phosphorus and potassium in branches of eight Mediterranean evergreens. *New Phytologist*. 2005;168:167-178.
38. Ostman NL and Weaver GT. Autumnal nutrient transfers by retranslocation, leaching, and litterfall fall in a chestnut oak forest in southern Illinois. *Can J Forest Res*. 1982;12:40-51.
39. Fife DN, et al. Retranslocation of foliar nutrients in evergreen tree species planted in a Mediterranean environment. *Tree physiol*. 2008;28:187-196.
40. Antonopoulos-Domis M, et al. Compartment model for long-term contamination prediction in deciduous fruit trees after a nuclear accident. *Health Phys*. 1990;58:737-741.
41. Antonopoulos-Domis M, et al. Long-term radiocesium contamination of fruit trees following the Chernobyl accident. *Health Phys*. 1996;71:910-914.
42. Likuku AS. The Influence of Topography and Vegetation, Canopy on the Deposition of Atmospheric Particulates Studied with ²¹⁰Pb and ¹³⁷Cs Soil Inventory Measurements. Ph.D. thesis, School of Physics University of Edinburgh. 2003;

43. Maathuis FJM and Sanders D. Mechanisms of potassium absorption by higher plant roots. *Physiol Plant*. 2006;96:158-168.
44. Ronneau C, et al. Radiocaesium and Potassium Behaviour in Forest Woody plants. *J Environ Radioact*. 1991;14:259-268.
45. Sombre L, et al. Long-term radiocesium behaviour in spruce and oakforests. *Sci Total Environ*. 1994;157:59-71.
46. Yoshida S, et al. Distribution of radiocesium and stable elements within a pine tree. *Radiat Prot Dosim*. 2011;146:326-329.
47. Zhu YG and Smolders E. Plant uptake of radiocaesium: A review of mechanisms, regulation and application. *J Exp Bot*. 2000;51:1635-1645.