

Waste Management: A Review

Divya Tamma^{1*} and Nitin Panwar²

¹Department of Biotechnology, Sheffield Hallam University, UK

²Department of Biotechnology, Graphic Era University, India

Review Article

Received: 27/09/2016
Accepted: 25/12/2016
Published: 30/12/2016

*For Correspondence

Divya Tamma, Department of Biotechnology, Sheffield Hallam University, UK, Tel: +91-9989239898.

E-mail:

divya.tamma88@gmail.com

Keywords: Organic and inorganic waste, Environment, Waste management, Biomass

ABSTRACT

Waste management is a new edge through which we can manage the miscellaneous type of waste generated throughout many industrialized and regular procedures. This article gives an ephemeral review on how to manage waste from the point of origin to its decomposition and also laid emphases on why there is a need for management of Organic and Inorganic waste and their adverse effects on environment.

INTRODUCTION

According to, Department for Economic and Social Information and Policy Analysis (USA), waste management is all the activities and actions required to manage waste from its inception to its final disposal [1]. The term waste management includes treatment of all kinds of waste, from its point of generation, extraction, processing of raw materials to their final product consumption. Waste is generally produced during human activities, municipal, agricultural or social activities, to avoid the environmental disturbances caused by their disposal in to nature; waste management is highly recommended.

During old days waste produced is of less percentage, because of the less population and minor use of natural resources. Waste was mostly from ashes produced from burnt things or slight social activities. The ashes and human biodegradable waste thus produced was released back in to land or atmosphere leaving minimum environmental impression. The household utensils are mostly made of wood or metal, which were passed on to future generations or reused [2-4]. The early 20th century set the industrial revolution followed by population growth around the world expanding the urban areas. This ultimately led to buildup of waste resulting rapid deterioration of sanitation levels and quality of life in metropolitan cities.

Benefits

With environment at risk one should consider that waste is not something that is disposed without concern of its consequences and future use. It can be served as valuable resource, if it is treated right and processed with good

practice. Globally waste management has wide range of benefits. From economic benefits to social, it has huge impact [5]. Those benefits include:

Economic benefits: Economy is considered important among all the factors of a country. Waste management can help in improving economy very efficiently, if waste is treated and recycled. With the natural resources at risk and in search of alternate energy resources, waste can be used to create markets for recycling and an energy source. Efficient practicing of this treated waste in production and consumption of valuable products can lead to recovered materials, new jobs, business opportunities and finally impacting social economic status.

Social: Health deterioration is the major adverse effect caused by environmental degradation. By continues practice of waste management, health risks are reduced, resulting better societies. Improved social advantages give rise to employment opportunities and uplifting of community from poverty [6].

Environmental: Waste disposed in to the environment is not only a harm to human but also to animal and plant kingdom. By reducing their adverse effects can provide better quality of air, water and reduction of greenhouse gas emissions. Reusing and recycling of waste can result in minimizing the exploration of natural resources and improvement of environment.

Inter-generational equity: Leaving a better planet for future generations is a human responsibility. With right waste management practices, future generations can be provided with better economy, more inclusive society and a cleaner environment.

INDUSTRIAL WASTE

Industrial waste is defined as a liquid, solid and gaseous wastes originating from the manufacture of specific products [4]. It is the mixture of several varieties of impurities and this reason alone makes its treatment, a special task. Considering its threat to environment, manufacturing companies started prioritizing the closed circuits and product recovery methods in various production processes. By introducing these methods to treat the waste, can contribute to the protection of eco-system and reducing the cost of production too. WABAG is the long-term experience in the industrial wastewater treatment sector. Mechanical, biological and chemical physical process is the steps followed during waste treatment. During biological treatment of waste water, it is exposed to anaerobic treatment, as it is considered as conventional, space saving and high performance process [7]. Depending on the characteristics of waste, several other processes are also implemented. Rapid industrialization led to growth of industrial sectors like sugar, pulp and paper, fruit and food processing, sago starch, distilleries, dairies, tanneries, slaughterhouses, poultries and many more [8]. These industries generate huge quantity of solid and liquid wastes. Regardless of the policies introduced by pollution control, waste produced is generally dumped on land or discharged in to water bodies, resulting in environmental pollution and health hazard.

Organic Chemicals

Pesticides, pharmaceuticals, paints, dyes, petroleum, detergents, plastics, paper pollution [9].

Tanning Industry

Economic status of any country is hugely influenced by its industrial production and marketing. Leather processing and tanning industry contributes hugely to economic activity all around the world. But processing of leather leaves an uncontrolled tannery effluent. When these effluents are released in to natural water bodies without prior treatment, can lead to environmental degradation and health hazard to human beings [10]. The tannery effluent produced from traditional or conventional leather processing contains a high concentration of organics (COD/BOD), Suspended Solids (S.S) and inorganics like NH₄-N, SO₄²⁻ /S₂⁻, Cr(III) and Chlorides [11-13]. These organic and

inorganic products from tannery effluent can cause health risks to aquatic ecosystem, human and environmental degradation [8]. Toxic chemicals in tanning effluents cause depletion of oxygen in water and affect the fishes, other aquatic flora and fauna [14].

Dyes: Water released from textile dye manufacturing and dyeing mills consist of large amounts of colored effluents [15]. Color is the first pollutant to be considered in wastewater [16] as it blocks the sunlight passing in to water. The estimation of total dye consumption worldwide is more than 107 kg/year [17]. Congo red (CR) is anionic dye that has benzene [18] and malachite green (MG) is a cationic dye [19]. In aqueous solution, anionic dyes carry a net negative charge due to the presence of sulphonate (SO_3^-) groups, while cationic dyes carry a net positive charge due to the presence of protonated amine or sulfur containing groups [20]. Dyes in industrial effluent can cause allergic dermatitis, skin irritation, dysfunction of kidney, liver, brain, reproductive and effects to central nervous system [21-22]. Due to presence of high color compounds, dyes are considered to be toxic and even carcinogenic [23-24]. Even in minute concentrations, these toxins widely affect the aquatic ecosystem and make water unacceptable for various household or agricultural purposes [25].

Treatment

- Various techniques like adsorption, nano-filtration, electro kinetic coagulation, coagulation and precipitation, advanced chemical oxidation, electrochemical oxidation, ozonation, liquid-liquid extraction and few other biological process can be employed for the removal of textile dyes from wastewaters [26]. Adsorption is the most promising and extensively used method for removal of both inorganic and organic pollutants from contaminated water [27]. Activated carbon have unique molecular structure, high porosity and an extensive surface area, this reason alone makes it an effective adsorption technique for dye removal [28].
- An organic pollutant from industrial waste is the reason for waste water being considered pollutant, adsorption by activated carbon is considered to be the best management for treating organic waste. Rhodamine B is a water soluble and basic red cationic xanthene class dye that is commonly found as tracer fluorescent. Activated carbon that is developed from coconut shell can be applied for removal of Rhodamine-B from wastewater successfully.
- The methylene blue number, iodine number and BET surface area of the prepared carbon were found to be 80 mg g⁻¹, 600 mg g⁻¹, and 1200 m² g⁻¹ respectively [29]. AE (2015) investigated that sodium dodecyle sulfate coated tea waste (SCTW) has excellent adsorption capacity for the removal of methylene blue from aqueous solutions. It can be concluded as a promising advanced adsorbent in environmental pollution clean-up.

Fossil Fuels

Power stations, coal-fired plants, lead, mercury, cadmium and chromium, as well as arsenic, selenium and nitrogen compounds. Fossil fuels are formed by natural processes such as anaerobic decomposition [30] of buried dead organisms, containing energy originating in ancient photosynthesis [31]. Fossil fuels contain high percentages of carbon, petroleum, coal, natural gas, kerosene and propane [32]. Fossil fuels contain volatile materials with low carbon: hydrogen ratios like methane, liquids like petroleum and nonvolatile materials like pure carbon and anthracite coal. Methane can be found in hydrocarbon fields either alone, associated with oil, or in the form of methane clathrates [33].

Crude oil: Oil wastes that are produced from crude oil companies, when dumped or burned without treating leave a huge risk to both ecosystems and human health and causes serious environmental consequences [34]. The chemical composition of oil sludge is complex and depends on the source. It is mainly composed of alkanes, aromatics,

asphaltenes and resin [35-38]. The common contamination caused by crude oil companies is spillage of its waste oil in to aquatic environment due to leakage in to water body, during the process of oil exploration and transportation [39]. This contamination cause disturbances to aquatic environment and is a big threat for the evolution of macrophyte [40]. The toxic chemicals present in the crude oil, depending on its water soluble fraction (WSF), can also be lethal in acute or chronic levels [41-43].

These toxins in crude oil carry specificity of finding their way in to the body system of aquatic animals (Fishes) through the gills, digestive tract and general body surface causing significant damage to the internal organs and tissues [44]. Research says that there is a relationship between contaminants exposure, various biological responses and changes to target organs of fish [45]. Some authors argue that histological alterations are sensitive tools that can be used to detect the effect of different toxins and different compounds on different organs, altogether as a good environmental stressor indicator for bioassay [46-47]. Fish exposed to pollutants [48] such as crude oil effluents cause histopathological changes in different tissues and organs such as gills, liver, kidney, spleen [49]. Depending on the stressor and the intensity of toxic agents, histological changes differ [50-53].

Automotive sector: Rapid industrialization and automobiles led to excessive use of fossil fuels and to serious environmental problems like climate change, deterioration of the ozone layer and acid rain. With natural resources at risk due to their extreme exploitation, new energy search has become a need. Thus biofuels is a great alternative [54]. Biodiesel that is produced from microalgae is considered as promising alternative because of its high growth rate, high capacity for lipid accumulation, CO₂ absorption capacity and ease of cultivation both outdoor tanks (raceway) and in closed reactors (photobioreactor) [55-57].

Alternative

- In search of alternative energies, industrialized countries started developing their primary energy fuel from renewable resources such as solar energy, wind energy, geothermal energy, biomass energy, hydropower, ocean energy and secondary energy source such as hydrogen energy.
- Proton Exchange Membrane Fuel Cell (PEMFC) is one of the alternative energy. In PEMFC, the chemical energy is converted directly into electricity and then to heat. It is estimated that PEMFC become an efficient and clean energy by 2020 [58]. One can conclude that automotive industry is developing, changing and playing a vital role in determining the fuel choice [59].
- Biogas is another alternative energy source. Biomass can efficiently produce methane and biogas when treated with fungal co-cultures of *Trichoderma viride*, *Aspergillus niger* and *Fusarium oxysporum*. This biological pretreatment is effective and cost reducing [60].
- Simarouba (*Simarouba glauca*) is commonly known as paradise tree. As per ASTM specification biodiesel, this paradise tree possesses important fuel properties [61]. Research says that automobile engine works smoothly on simarouba methyl ester when compared to diesel operation. Thus simarouba biodiesel can be successfully substituted as alternative fuel for CI engine.
- To reduce the effects of fossil fuel utilization and of economic interest, there is continues interest in search of renewable energy. Animal fats and vegetable oils can produce ethanol that can be used for primary renewable transportation [62]. The fatty acid methyl esters from animal fats and vegetable oils, referred to as biodiesel, can provide significant reductions in particulate matter, CO and HC emissions [63]. Alternate fuels which are extracted from vegetable oils will positively reduce the usage of fossil fuels. Oil extracted from orange peel and

cotton seed can be used as alternate energy when it is blended with petrol and used in petrol Engines. Advantage is no modification of engine is needed for this application [64].

FOOD AND AGRICULTURE

Though waste from food and agriculture is considered biodegradable and non-toxic, it has high concentrations of biochemical oxygen demand (BOD) [65-67] and suspended solids (SS).

The differences in BOD and pH in effluents from vegetable, fruit, and meat products, seasonal nature of food processing and post-harvesting, makes food and agriculture waste complex to treat. According the Natural Resources Defense Council [68] more than 40% of the food in the United States is wasted during crop production, transportation and final consumption, which accounts for \$165 billion, each year go to trash. Produced food waste ends up in landfills and release methane in to the atmosphere [69] which is another form of air pollution.

Treatment

- Food wastes are rich in organic matter and during degradation, produce methane, a greenhouse gas. This waste can be utilized to produce biogas, under conditions like feedstock characteristics reactor design and anaerobic digestion process. To achieve food waste treatment and its conversion into biogas two steps are carried out, one in single phase and other in high-rate two-phase anaerobic digestion [70-73].
- The organic matter from agricultural crops and wastes, animal wastes, forest and wood residues, plants and municipal waste is called biomass and is stored as chemical energy. This energy can be released as biogas such as methane (CH₄), hydrogen (H₂) and carbon dioxide (CO₂) through the anaerobic digestion process [74-77].
- Sugarcane waste-Press mud can be utilized for biogas production. When industrial waste is subjected to anaerobic digestion in a laboratory large scale floating drum bio digester with two different conditions, biogas yield is obtained from dry pressmud and methane concentration (CH₄) can be reached to 67% compared to wet pressmud [78]. Hence industrial waste (pressmud) is potential source for energy production.

MINING

Recovery of ores like metals, coal, oil shale, gemstones, limestone, dimension stone, rock salt, potash, gravel, and clay.

Iron

A powerful reduction reaction in blast furnaces is applied during production of iron from its ores and water is used for cooling, during this process the effluent produced is inevitably contaminated with ammonia and cyanide [79]. Production of coke from coal in coking plants and by-products separation include water cooling process. Benzene, naphthalene, anthracene, cyanide, ammonia, phenols, cresols and wide range of more complex organic compounds known collectively as polycyclic aromatic hydrocarbons (PAH) from these industry effluents are considered carcinogenic.

Heavy Metals

Population growth directly affected increased need for natural resources and its exploitation has become an important economic activity worldwide. But adverse effects are hugely due to its poor exploitation processes and wrong disposal of mine tailings [80-81]. Several researchers say that heavy metals are bio concentrated or bio accumulated in one or several compartments across food webs [82-83]. This contamination of natural ecosystem with trace of toxic elements directly affects the natural functioning of aquatic organisms and indirectly become reason

for decrease in biodiversity and extinction of sensitive taxa [84]. Metal bioaccumulation is considered important in regard to human health, especially when humans consume the accumulators. This phenomenon is considered important in assessing environmental quality and chemical survey of water and sediment [85]. Heavy metals enter the aquatic environment mainly by anthropogenic sources. Fish is at the top of the aquatic food chain, and during its life can accumulate large amounts of toxic elements [86-89]. Another form of heavy metal contamination is in residential areas, where most of Pb contamination is attributed to the deterioration of lead-based paint for housing [90].

Toxins

Chronic exposure to benzene at minimum concentrations can cause leukemia, mercury and cyclodienes, kidney damage and sometimes these effects are irreversible. PCBs and cyclodienes leave toxic effects on liver. Organophosphates and carbamates on chronic level induce a chain of responses leading to neuromuscular blockage. Chlorinated solvents induce liver changes, kidney changes and depression of the central nervous system [91-94]. Toxins carrying heavy metals can cause headache, nausea, fatigue, eye irritation and skin rash.

Treatment

- A new technique called Phytoremediation is developed recently, it is considered as an effective technology for treating heavy metals. It uses plants to degrade, assimilate, metabolize, or detoxify metals, hydrocarbons, pesticides and chlorinated solvents. To treat a variety of hazardous chemicals, phytoremediation is a best approach and it is cost-effective and resource-conservative technique [95].
- Scolecite natural zeolite is capable of removing metal ions Ni²⁺, Pb²⁺, Zn²⁺, Cd²⁺, Fe³⁺, Cr³⁺ from industrial waste water samples. The percentage of removal of heavy metal ions by scolecite is 95 % and 99.9% of Ni²⁺, Pb²⁺, Zn²⁺, Cd²⁺, Fe³⁺ and Cr³⁺ can be removed. Thus natural zeolite can be used effectively for the removal of these metal ions from industrial wastewater [96]. This naturally occurring material provides a substitute for the use of other materials as adsorbent due to its availability and low cost. Thus treated industrial waste water can be reused for washing, irrigation etc.

PULP AND PAPER INDUSTRY

Furans, phenols, insecticides

Wool

During Wool processing, flees are treated with water and thus produced water get contaminated with animal fats and chemicals from insecticides.

Paper

The industry of pulp and paper produces effluent that is high in suspended solids and BOD [97]. Plants that bleach wood pulp for paper making may generate chloroform, dioxins, furans, phenols and chemical oxygen demand (COD). Pulp and paper mills utilize huge amount of lingo-cellulosic components of plants and chemicals during their manufacturing and considered as polluting industries because of huge amount of waste material that is released into the environment [98].

Pulp

It is produced by 40-50% of plant materials, which are heavily with organic material. Thus produced effluent composes of compounds like chlorinated lingsulphonic acid, chlorinated resin acid, chlorinated phenol and chlorinated hydrocarbon and many more organic chemicals. As a result of different processes applied in wood and pulp bleaching, this industry discharges large volumes of brown colored effluents that are diverse in nature [99-100].

Due to their high chemical diversity in nature, these pollutants cause clastogenic, carcinogenic and mutagenic effects on fish and other aquatic communities in recipient water bodies ^[101].

Treatment

- The best possible way of treating effluents from paper mill plants is primary clarification succeeded by secondary treatment, generally of a biological nature. There are numerous biological treatment systems available and the most common is the activated sludge process ^[102-103].
- Activated sludge system technology is a process where agitation of the effluent is achieved, in the presence of aerobic bacteria, protozoa, metazoa and atmospheric oxygen for a sufficient period to metabolize and to flocculate a large part of the organic material.
- Protozoa play a secondary but important role in wastewater system purification ^[104-105]. The protozoa in the activated sludge treatment process are of four main classes: amoebae, flagellates and ciliates and metazoa that are separated into rotifers, nematodes and Oligotrichia such as Aelosoma ^[106]. Industrial effluents composing of specific characteristics are treated with activated sludge process and depending on specifications, kind of protozoa are used ^[107-108].
- Phenol is an organic compound that is widely used in petrochemical, oil refining, plastic, leather pharmaceutical and pesticide industries. Papita Das et.al, (2015), suggested that soil adsorbent can be implemented as an efficient liner material for the removal of phenol and phenolic compounds from wastewater ^[109].

CONCLUSION

Rapid industrialization, population growth and many other social factors contributed to the increase in production of waste and environmental disturbances caused by them. Consequently these changes in environment are affecting whole food chain. This review article laid emphasis on treatment of waste produced from industries or household, before its disposal in to nature, either water bodies or land. Thus treated waste is economically and socially beneficial to the humans and their environment. The organic wastes like dyes can be reused after treatment and they are not hazardous to the habitat if released after adsorption of harmful chemicals. Instead of using fossil fuels, the renewable energy resources can be brought into use. The agricultural waste can be minimized by using them as biodegradable sources of energy production. Phytoremediation is a new technology that minimizes the waste rendered from mining. Considering the responsibility to leave a better planet for next generations, this exploitation of natural resources should be minimized to a great extent. By reusing and recycling the waste, one can reduce the adverse effects on environment.

REFERENCES

1. Glossary of Environment Statistics: Series F, No. 67 / Department for Economic and Social Information and Policy Analysis, United Nations. New York: UN, 1997.
2. Szalata L, et al. Environmental Management Systems in Municipal Waste Management, On the Example of a Selected Waste Disposal Plant. J Pet Environ Biotechnol. 2016;7: 294.
3. Mickael D Categorization and Sorting for Waste Management. Int J Waste Resour. 2016;6:227.
4. Kamsano NS, et al. Waste-to-Wealth: Bio-Recycling Centre as Living Laboratory Element to Create Integrated Bio-Waste Management for Institution. Int J Waste Resour. 2016;6:217.
5. <http://www.autoregulations.ie/compliance/legal-responsibility/waste-management>

6. Mbu AD. The Influence of Environmental Awareness on Human Attitude to Solid Waste Management in Boki Local Government Area of Cross River State. *J Pollut Eff Cont.* 2015;3:144.
7. Batham M, et al. Time Efficient Co-composting of Water Hyacinth and Industrial Wastes by Microbial Degradation and Subsequent Vermicomposting. *J Bioremed Biodeg.* 2014;5:222.
8. Papargyropoulou E, et al. Towards Sustainable Resource and Waste Management in Developing Countries: The Role of Commercial and Food Waste in Malaysia. *Int J Waste Resources.* 2014;4:151
9. Kashiwaya, et al. "Tannery Wastewater Treatment by the Oxygen Activated Sludge Process". *Journal (Water Pollution Control Federation)*. Alexandria, VA: Water Environment Federation. 1980;52 (5): 999–1007. JSTOR 25040825.
10. Sabumon PC. Perspectives on Biological Treatment of Tannery Effluent. *Adv Recycling Waste Manag.* 2016;1:104.
11. UNIDO. Pollutants in tannery effluents, Regional Programme for Pollution Control in the Tanning Industry in South-East Asia, The Scope for Decreasing Pollution Load in Leather Processing. 2000.
12. Kaul SN, et al. Wastemangement in tanneries: Experience and outlook. *Journal of Indian Association of Environmental Management.* 2001;28: 56-76.
13. Boshoff G, et al. Tannery effluent as a carbon source for biological sulphate reduction. *Water Research.* 2004;38: 2651-2658.
14. Kongjiao S, et al. Simultaneous removal of organic and inorganic pollutants in tannery wastewater using electro coagulation technique. *Korean Journal of Chemical Engineering.* 2008; 25: 703-709.
15. Kolomaznik K, et al. Leather waste potential threat to human health and a new technology of its treatment. *Journal of Hazardous Materials.* 2008;160: 514-520.
16. Tan LS, et al. Adsorption of Textile Dye from Aqueous Solution on Pretreated Mangrove Bark, an Agricultural Waste: Equilibrium and Kinetic Studies. *Journal of Applied Sciences in Environmental Sanitation.* 2010;5:283-294.
17. Rangabhashiyam S, et al. Sequestration of Dye from Textile Industry Wastewater Using Agricultural Waste Products as Adsorbents. *Journal of Environmental Chemical Engineering.* 2013;1: 629-641.
18. Mittal A, et al. Freundlich and Langmuir Adsorption Isotherms and Kinetics for the Removal of Tartrazine from Aqueous Solutions Using Hen Feathers. *Journal of Hazardous Materials.* 2007;146: 243-248.
19. Sivakumar V, et al. Removal of Congo Red Dye Using an Adsorbent Prepared from *Martynia annua* L. Seeds. *American Chemical Science Journal.* 2014;4: 424-442.
20. Ahmad R, et al. Adsorption Studies of Hazardous Malachite Green onto Treated Ginger Waste. *Journal of Environmental Management.* 2010;91: 1032-1038.
21. Abbas A, et al. Comparative Study of Adsorptive Removal of Congo Red and Brilliant Green Dyes from Water Using Peanut Shell. *Middle-East Journal of Scientific Research.* 2012;11: 828-832.
22. Igwegbe, CA., et al. Adsorptive Treatment of Textile Wastewater Using Activated Carbon Produced from *Mucuna pruriens* Seed Shells. *World Journal of Engineering and Technology.* 2016; 4: 21-37.
23. Salleh MAM, et al. Cationic and Anionic Dye Adsorption by Agricultural Solid Wastes: A Comprehensive Review. *Desalination.* 2011;280: 1-13.
24. G.Crini. Non-conventional low-cost adsorbents for dye removal: a review. *Bioresour Technol.* 2006; 97:1061-1085.

25. Chen S, et al. Equilibrium and kinetics studies of methyl orange and methyl violet adsorption on activated carbon derived from phragmitesaustralis. *Desalination*. 2010;252: 149-156.
26. Maurya NS, et al. Evaluation of Adsorption Potential of Adsorbents: A Case of Uptake of Cationic Dyes. *Journal of Environmental Biology*. 2008;29: 31-36.
27. Mahmoud AS, et al. Removal of Dye from Textile Wastewater Using Plant Oils under Different pH and Temperature Conditions. *American Journal of Environmental Science*. 2007; 3: 205-218.
28. Prasad AL and Santhi T. Adsorption of Hazardous Cationic Dyes from Aqueous Solution onto Acacia nilotica Leaves as an Eco Friendly Adsorbent. *Sustainable Environmental Research*. 2012;22: 113-122.
29. Isah UA and Gatawa AI. A Kinetic Study of the Adsorption of Reactive Yellow 21 Dye on Flamboyant Shells Activated Carbon. *Advances in Applied Science Research*. 2012;3: 4036-4040.
30. Balasubramani K and Sivarajasekar N. Adsorption Studies of Organic Pollutants onto Activated Carbon. *International Journal of Innovative Research in Science, Engineering and Technology*. 2014;4:2.
31. Koyama and Tadashiro. "Gaseous metabolism in lake sediments and paddy soils and the production of atmospheric methane and hydrogen". *Journal of Geophysical Research*. 1963;68: 3971–3973.
32. Ivlev AA . Global Redox Carbon Cycle and Photosynthesis Development. *J Ecosys Ecograph*. 2016; S5:003.
33. "Gasoline as Fuel – History of Word Gasoline – Gasolin and Petroleum Origins". *Alternativefuels.about.com*. 2013-07-12. Retrieved 2013-08-27.
34. Sato Motoaki, *Thermochemistry of the formation of fossil fuels*. The Geochemical Society. Special Publication No.2. 1990.
35. Baheri H and Meysami P. Feasibility of fungi bioaugmentation in composting a flare pit soil. *J Hazard Mater*. 2002;89: 279-286.
36. Kurosawa K, et al. Triacylglycerol Production from Corn Stover Using a Xylose-Fermenting *Rhodococcus opacus* Strain for Lignocellulosic Biofuels. *J Microb Biochem Technol*. 2014;6: 254-259.
37. Lewis Liu Z and Wang X. A Reference Model System of Industrial Yeast *Saccharomyces cerevisiae* is needed for Development of the Next-Generation Biocatalyst toward Advanced Biofuels Production. *J Microb Biochem Technol*. 2015;7:e125.
38. Saldivar RP, et al. Algae Biofuels Production Processes, Carbon Dioxide Fixation and Biorefinery Concept. *J Pet Environ Biotechnol*. 2014;5:185.
39. Diallo M, et al. Thermodynamic properties of asphaltene: A predictive approach based on computer assisted structure elucidated and atomistic simulations. In: Yen TF, ChilingarianGV (eds) *Asphaltene and asphalts II, developments in petroleum science 40B*. Elsevier Amsterdam. 2000;103-127.
40. Mariano JB and LaRovere EL. Environmental impacts of the oil industry. *Petroleum engineering. Downstream*. 1999.
41. Vasyunina Ye A, et al. Estimation of Toxicity and Genotoxicity of Water, Bottom Sediments and Submerged Macrophyte *Elodea canadensis* of the Yenisei River in the Presence or Absence of Americium-241. *J Environ Anal Toxicol*. 2016; 6:389.
42. Jadoon S, et al. Effects of Crude Oil Contamination under the Controlled Conditions on the Physicochemical Properties of Water in Khurmala and Guwayar, Kurdistan Region, Iraq. *J Pollut Eff Cont*. 2016; 4:165

43. Saborimanesh N and Mulligan CN. Effect of Sophorolipid Biosurfactant on Oil Biodegradation by the Natural Oil-Degrading Bacteria on the Weathered Biodiesel, Diesel and Light Crude Oil. *J Bioremed Biodeg.* 2015; 6:314.
44. Ebonwu BI and Ugwu LLC. Effect of Crude Oil Water Soluble Fraction Toxicity on *Tilapia Guineensis* Fingerlings Using Histology of the Kidney as a Bioassay Indicator. *J Pet Environ Biotechnol.* 2016;7: 287.
45. Pathan TS, et al. Histological changes in the gill of freshwater fish, *Rasboradaniconius* exposed to paper mill effluent. *Iranican Journal of Energy & Environment.* 2010;1: 170-175.
46. Schwaiger J, et al. Effects of sublethal concentrations of triphenyltinacetate on rainbow trout (*Oncorhynchus mykiss*). *Arch Environ Contamination Tox.* 1996;30: 327-334.
47. Mallat J. Fish gill structural changes induced by toxicants and other irritants: A statistical review. *Can J Fish Aquat Sci.* 1995;42: 630-648.
48. Hagerty CL and Ramseur J. Deepwater horizon oil spill: Selected Issues for 1999. *Congressional Research.* 2010.
49. Khan F, et al. In Silico Approach for the Bioremediation of Toxic Pollutants. *J Phylogenetics Evol Biol.* 2013;4:161.
50. Bernet D, et al. Histopathology in fish: proposal for a protocol for assessment of aquatic pollution. *J Fish Dis.* 1999;22: 25-34.
51. El-Hussein and Marzouk A. Characterization of Petroleum Crude Oils using Laser Induced Fluorescence. *J Pet Environ Biotechnol.* 2015;6:240
52. Ipeaiyeda AR, et al. Biodegradation of Polycyclic Aromatic Hydrocarbons in Agricultural Soil Contaminated with Crude Oil from Nigeria Refinery using *Pleurotus sajor-caju*. *J Bioremed Biodeg.* 2015; 6: 301.
53. El Mahdi AM, et al. Performance of Isolated *Kocuria* sp. SAR1 in Light Crude Oil Biodegradation. *J Bioremed Biodeg.* 2015; 6:303.
54. Pérez L. Biofuels from Microalgae, A Promising Alternative. *Pharm Anal Chem Open Access.* 2016;2:e103.
55. Wei W. Anaerobic Co-digestion of Biomass for Methane Production: Recent Research Achievements. 2013; 1-10.
56. As'ad AM, et al. Solvent Dewaxing of Heavy Crude Oil with Methyl Ethyl Ketone. *J Pet Environ Biotechnol.* 2015;6:213.
57. Oje Obinna A, et al. Variation in the Carbon (C), Phosphorus (P) and Nitrogen (N) Utilization during the Biodegradation of Crude Oil in Soil. *J Pet Environ Biotechnol.* 2015;6:206.
58. Zhang B and Shahbazi A. Recent Developments in Pretreatment Technologies for Production of Lignocellulosic Biofuels. *J Phylogenetics Evol Biol.* 2011;2:108.
59. Taner T. Alternative Energy of the Future: A Technical Note of PEM Fuel Cell Water Management. *J Fundam Renewable Energy Appl.* 2015;5:163.
60. UK. <http://www.omicsonline.org/proceedings/the-role-of-biofuels-of-future-global-transport-mix-49910.html>
61. <http://www.omicsonline.org/proceedings/biological-wheat-straw-pretreatment-novel-fungal-cocultures-towards-sustainable-future-in-biofuel-production-50020.html>

62. Stanislav GB. A Novel General Methodology for Ribozyme-Mimetic Synthesis of Methyl Esters of Various Natural Amino Acids, Simulating the Prebiotic Biomolecule Creation. *J Chem Eng Process Technol.* 2015; 6:258.
63. Praveen AH. et al. Biodiesel as an Alternative Fuel for CI Engine: Review. *International Journal of Innovative Research in Science, Engineering and Technology.*
64. Muneeswaran R and Thansekha MR. Effect of Nox Emissions in a Bio-Fuelled Di Diesel Engine. *International Journal of Innovative Research in Science, Engineering and Technology.*
65. Manimaran P, et al. Experimental investigation of the orange peel oil and cotton seed oil blend with petrol as an alternate fuel for petrol engines. *International Journal of Innovative Research in Science, Engineering and Technology.* 2014;3:3.
66. Ho HL and Lau LY. Bioprocessing of Agricultural Wastes as Optimised Carbon Source and Optimisation of Growth Conditions for Xylanase Production by *Aspergillus Brasiliensis* in Agitated Solid State Fermentation (Ssf). *J Biodivers Biopros Dev.* 2014;1:125.
67. Ho HL. Xylanase Production by *Bacillus subtilis* Using Carbon Source of Inexpensive Agricultural Wastes in Two Different Approaches of Submerged Fermentation (SmF) and Solid State Fermentation (SsF). *J Food Process Technol.* 2015; 6:437.
68. <http://www.omicsonline.org/proceedings/processing-urban-waste-for-energy-for-reduced-greenhouse-gases-38978.html>
69. Gunders D. Wasted: How America Is Losing Up to 40 Percent of Its Food from Farm to Fork to Landfill. *Natural Resources Defense Council, NRDC Issue.* 2012.
70. Demirel B, et al. Production of Methane and Hydrogen from Biomass through Conventional and High-Rate Anaerobic Digestion Processes. *Critical Reviews in Environmental Science and Technology.* 2010;40: 116-146.
71. Babu MD, et al. Development and Validation of a GC-MS with SIM Method for the Determination of Trace Levels of Methane Sulfonyl Chloride as an Impurity in Itraconazole API. *J Anal Bioanal Tech.* 2016; 7:316
72. Shehu H, et al. Study of the Selectivity of Methane over Carbon Dioxide Using Composite Inorganic Membranes for Natural Gas Processing. *J Adv Chem Eng.* 2016; 6:150.
73. Plocoste T, et al. Effect of Leachate Recirculation on Landfill Methane Production in a Tropical Insular Area. *Innov Ener Res.* 2016;5:138.
74. Yogeshvari JK, et al. Rapid Methods for Isolation and Screening of Methane Degrading Bacteria. *J Bioremed Biodeg.* 2016;7:322.
75. Shu CH, et al. Improving Biodegradation of Rice Straw Using Alkaline and *Aspergillus niger* Pretreatment for Methane Production by Anaerobic Co-Digestion. *J Bioprocess Biotech.* 2015;5:256
76. Rajput JD, et al. Design, Synthesis and Biological Evaluation of Novel Class Diindolyl Methanes (DIMs) Derived from Naturally Occurring Phenolic Monoterpenoids. *Med chem (Los Angeles).* 2016;6:123-128.
77. Plocoste T, et al. Estimation of Methane Emission from a Waste Dome in a Tropical Insular Area. *Int J Waste Resour.* 2016;6:211.
78. Sreekanth P. Outlook on Anaerobic Digestion Process and Methods to Evaluate Biogas A. *Rese & Rev: Jour Ecol Environ Scien.* 2016.

79. Sathish S and Vivekanandan S. Experimental Investigation on Biogas Production Using Industrial Waste (PressMud) To Generate Renewable Energy. *Int J Inn Res Sci, Engi Tech.* 2015;4:2.
80. Jump up EPA. "Section 7: Wastewater Characterization." Development Document for Final Effluent Limitations Guidelines and Standards for the Iron and Steel Manufacturing Point Source Category. 2002; Document no. EPA-821-R-02-004. pp. 7-1ff.
81. Stael C and Cumbal L .Optimized Synthesis of Multicomponent Nanoparticles for Removing Heavy Metals from Artificial Mine Tailings. *Biol Med (Aligarh).* 2016;8: 288.
82. Peacey V and Yanful EK. Metal Mine Tailings and Sludge Co-Deposition in a Tailings Pond. *Water, Air, and Soil Pollution.* 2003;145: 307-339.
83. Soegiarto A and Irawan B. Bioaccumulation of heavy metals in aquatic animals collected from coastal waters of Gecko Indonesia. *J Water Env Pollu.* 2009;2: 95-100.
84. Celechovska O, et al. Entry of heavy metals into food chains: A 20-year comparison study in northern Moravia (Czech Republic). *Acta Veterinaria Brno.* 2008;77: 645-652.
85. Bonanno G and Lo Giudice RL. Heavy metal bioaccumulation by the organs of *Phragmites australis* (common reed) and their potential use as contamination indicators. *Ecological Indicators.* 2010;10: 639-645.
86. Javanshir A, et al. Influence of Water hardness (Calcium concentration) on the absorption of Cadmium by the mangrove oyster *Crassostrea gasar* (Ostreidae; Bivalvia). *J Food Agri Env.* 2011;9: 724-727.
87. Fatema K, et al. Determination of Toxic Metal Accumulation in Shrimps by Atomic Absorption Spectrometry (AAS). *J Environ Anal Chem.* 2015; 2:140.
88. Saha B. Toxic Metals and Plants. *J Plant Biochem Physiol.* 2015; 3:e129.
89. Rolli NM, et al. Metal Accumulation Profile in Roadside Soils, Grass and *Caesalpinia* Plant Leaves: Bioindicators. *J Environ Anal Toxicol.* 2015;5:319.
90. Nwabunike MO. The Effects of Bioaccumulation of Heavy Metals on Fish Fin Over Two Years. *J Fisheries Livest Prod.* 2016;4:170.
91. Skaf DW, et al. The Effects of Photocatalyst and Solution Co-Contaminants on Photocatalytic Oxidation of 1,3-Dinitrobenzene in Aqueous Semiconductor Oxide Suspensions. *J Chem Eng Process Technol.* 2016;7:275.
92. Trivedi MK, et al. Characterization of Biofield Energy Treated 3-Chloronitrobenzene: Physical, Thermal, and Spectroscopic Studies. *J Waste Resources.* 2015;5:183.
93. Fares F, et al. An Innovative Complex of Benzene-Poly-Carboxylic Acid and Molybdenum, for Prevention and Treatment of Radiation Dermatitis. *Med chem.* 2015; 5:447-451.
94. Ha H, et al. Analysis of Pollution Hazard Intensity: A Spatial Epidemiology Case Study of Soil Pb Contamination. *Int. J. Environ. Res. Public Health.* 2016;13: 915.
95. Joseph TUR, Ramesh KB. Heavy Metal Risk Assessment in Bhavanapadu Creek Using Three Potamidid Snails - *Telescopium telescopium*, *Cerithidea obtusa* and *Cerithidea cingulata*. *J Environ Anal Toxicol.* 2016; 6:385.

96. Annie Melinda Paz-Alberto, Jose Lorenzo D. Vizmonte, Gilbert C. Sigua. Assessing Diversity and Phytoremediation Potential of Mangroves for Copper Contamination Sediments in Subic Bay, Philippines. *International J Plant animal env Sci.* 2015;5:4.
97. Almalih MA, et al. Removal of Heavy Metal Ions from Industrial Wastewater by *Scolecite*. *J Environ Anal Toxicol.* 2015;5:302.
98. EPA. "Permit Guidance Document: Pulp, Paper and Paperboard Manufacturing Point Source Category." 2000; Document no. EPA-821-B-00-003. pp. 4-1ff.
99. Kumar R, et al. Characterization and Isolation of Fungi for Removal of Color from Pulp and Paper Mill Effluent, Meerut (India). *J Environ Anal Toxicol.* 2015;5:324.
100. Durán N, et al. A new alternative process for Kraft E1 effluent treatment. *Biodegradation.* 1994; 5:13-19.
101. Moraes SG, et al. Remediation of Kraft E1 and black liquor effluents by biological and chemical processes. *Environmental Chemistry Letters.* 2006;4: 87-91.
102. Freire RS, et al. Remediation and toxicity removal from Kraft E1 paper mill effluent by ozonation. *Environ Technol.* 2001;22: 897-904.
103. Thompson G, et al. The treatment of pulp and paper mill effluent: a review. *Bioresour Technol.* 2001;77: 275-286.
104. Assalin MR, et al. Combined system of activated sludge and ozonation for the treatment of kraft E1 effluent. *Int J Environ Res Public Health.* 2009;6: 1145-1154.
105. Atta NN, et al. Anaerobic Co-Digestion of Wastewater Activated Sludge and Rice Straw in Batch and Semi Continuous Modes. *J Fundam Renewable Energy Appl.* 2016;6: 204.
106. Madoni P. Microfauna Biomass in Activated Sludge and Biofilm. *Water Sci Technol.* 1994;29: 63-66.
107. Tilley E, et al. *Compendium of Sanitation Systems and Technologies.* Duebendorf, Switzerland: Swiss Federal Institute of Aquatic Science and Technology (EAWAG) and Water Supply and Sanitation Collaborative Council (WSSCC). 2008.
108. Amaral AL, et al. Survey of Protozoa and Metazoa populations in wastewater treatment plants by image analysis and discriminant analysis. *Environmetrics.* 2004;15: 381-390.
109. Buitrón G and González A. Characterization of the microorganisms from an acclimated activated sludge degrading phenolic compounds. *Water Sci Technol.* 1996;34: 289-294.
110. Das P, et al. Phenol Adsorption onto Various Soil Composite Membranes: Insight into Process Kinetics, Modelling and Optimisation Using Response Surface Methodology. *Hydrol Current Res.* 2015; 6:203.