# A Brief Overview on Physicochemical Properties in Medicinal Chemistry

#### **Blanche Hurlbutt\***

Department of Chemistry, Harvard University, Boston, USA

#### Editorial

Received: 04-Apr-2022, Manuscript

No,JOMC-22-63159;

Editor assigned: 07-Apr-2022, PreQC No. JOMC-22-63159(PQ); Reviewed: 18-Apr-2022, QC No.

JOMC-22-63159:

Revised: 25-Apr-2022, QC No.

JOMC-22-63159

Published: 04-May-2022, DOI:

10.4172/j.med.orgnichem.9.2.005.

#### \*For Correspondence:

Blanche Hurlbutt, Department of Chemistry, Harvard University, Boston, USA

E-mail: blancheh12@gmail.com

#### **ABOUT THE STUDY**

Medicinal chemistry is a discipline at the intersection of chemistry, particularly synthetic organic chemistry, and pharmacology and other biological specialties, where they are involved in the design, chemical synthesis, and development of pharmaceutical agents, or bio-active molecules, for the market. In particular, medicinal chemistry in its most common form focusing on small organic molecules includes synthetic organic chemistry and aspects of natural products, as well as computational chemistry in close collaboration with chemical biology, enzymology, and structural biology, all with the goal of discovering and developing new therapeutic agents. In practise, it entails chemical aspects of identification, followed by systematic, thorough synthetic modification of new chemical entities to make them suitable for therapeutic use [1].

It encompasses the synthetic and computational aspects of studying existing drugs and agents in development in relation to their bioactivities i.e., understanding their structure activity relationships. Pharmaceutical chemistry is concerned with the quality of medicines and the fitness of medicinal products for their intended use. Additional chemistry and analysis are required, first to identify triage compounds that do not provide series with suitable SAR and chemical characteristics associated with long-term potential for development, and then to improve remaining in phrases of the desired primary activity, as well as secondary activities and physiochemical properties, so that the agent will be useful when administered in real patients. In this regard, chemical modifications can improve the

candidate compounds' recognition and binding geometries and thus their affinities for their targets, as well as the physicochemical properties of the molecule, which underpin necessary pharmacokinetic/pharmacodynamic and toxicologic profiles. The final stages of synthetic chemistry involve the production of a lead compound in sufficient quantity and quality to allow for large-scale animal testing, followed by human clinical trials. This includes optimising the synthetic route for bulk industrial production and identifying the best drug formulation [2]. The former is still the domain of medicinal chemistry, while the latter includes the specialization of formulation science. Process synthesis is a synthetic chemistry specialization in medicinal chemistry that aims to adapt and optimize the synthetic route for industrial scale syntheses of hundreds of kilogrammes or more. It entails thorough knowledge of acceptable synthetic practice in the context of large scale reactions. Pharmaceutical structures are evaluated in a variety of ways, including as a means of predicting efficacy, stability, and accessibility [3]. Lipinski's rule of five considers the number of hydrogen bond donors and acceptors, the number of rotatable bonds, the surface area, and the lipophilicity [4]. Other criteria used by medicinal chemists to evaluate or classify their compounds include synthetic complexity, chirality, flatness, and aromatic ring count. Prior to ligand synthesis, structural analysis of lead compounds is frequently performed using computational methods. This is done for a variety of reasons, including, but not limited to, time and financial concerns. After the ligand of interest has been synthesized in the laboratory, traditional methods of analysis are used [5]. Medicinal chemistry is an interdisciplinary science by definition, and practitioners must have a strong background in organic chemistry that is eventually combined with a broad understanding of biological concepts related to cellular drug targets [6]. Scientists in medicinal chemistry are primarily industrial scientists, working as part of an interdisciplinary team that uses their chemistry abilities, particularly their synthetic abilities, to design effective therapeutic agents based on chemical principles [7]. Bioisosteres are chemical substituents or groups with similar physical or chemical properties that produce biological properties that are broadly similar to another chemical compound. In drug design, the goal of exchanging one bioisostere for another is to improve a compound's desired biological or physical properties without causing significant changes in its chemical structure. The main application of this term and its techniques is in pharmaceutical sciences [8]. Bioisosterism is used to reduce toxicity, change bioavailability, or modify the activity of a lead compound, and it may also change the lead's metabolism. Symbioimine, an amphoteric minimum metabolite from the din flagellate Symbiodinium sp., represses osteoclast separation. Other novel alkaloids, for example, pinnamine, pinnaic acids, halichlorine, and zamamistatin, are likewise depicted. Numerous sorts of bioactive nitrogenous mixtures, like peptides, indols, oxazoles, and thiazoles, have been distinguished from marine spineless creatures. The genuine beginnings or forebears of these metabolites have been proposed to be microorganisms, i.e, microalgae, microscopic organisms, and growths. These microorganisms are brought through beneficial interaction, affiliation, a pecking order, and different types of supplement reliance with have creatures. Thusly, the separation of bioactive metabolites from refined marine microorganisms, like advantageous din flagellates and microbes, as well as from their host creatures, has been all around explored. A few alkaloid metabolites disconnected from cyanobacteria have been recommended to assist with inhibiting predation by marine herbivores, like fish and ocean imps. In any case, the genuine job of most marine bioactive alkaloids in the environment has not been very much explained. In our continuous quest for bioactive metabolites from marine organic entities, a few novel heterocyclic alkaloids, for example, Pinnatoxins, norzoanthamine, pinnaic acids, zamamistatin, and symbioimine, have been disconnected. This work includes the designs, natural exercises, and biogenesis of these

bioactive heterocyclic marine alkaloids, alongside forward-thinking themes. Alkaloids are nitrogen-containing intensifies that happen normally in plants as well as in microorganisms, marine living beings. Numerous sorts of alkaloids with unprecedented designs and critical natural exercises have been segregated from marine living beings [9.10]. They keep on giving lead structures in the quest for new medications or natural tests for physiological examinations. As new and more confounded sicknesses are experienced around the world, the significance of novel bioactive alkaloids has expanded because of their expected application in chemotherapy. The presence of N-O single bonds is by and large considered as an underlying component that presents genuine dangers to the medication similarity of the subsequent atom. This cautioning gets from the overall perception that the majority of the medications bearing amino-or nitro-moieties produce receptive N-OH metabolites. These N-OH metabolites are generally produced by biochemical change through hydroxylation or oxidation of amines or by decrease of nitro bunches present in the parent sedates and are viewed as harmful metabolites. These receptive metabolites can either covalently tie to nucleic acids or collaborate with proteins and, hence, be liable for quite a long time impacts for example cell putrefaction, touch and blood dyscrasia. The high reactivity of these N-OH metabolites towards nucleophilic species is for the most part advanced by an underlying formation with inorganic sulphate to deliver comparing ester, which further ionizes to create the electrophilic nitrenium species. These cationic species covalently tie to nucleic acids as well as other destinations of cell parts, to create stable adducts which produce serious harmfulness and, at last, may prompt the development of a threatening tumours. There is a developing interest in growing the synthetic space of new possible restorative specialists and might address a recently arising underlying theme.

#### **REFERENCES**

- 1. Sakariyau A, et al. An investigation on secondary school students' attitude towards science in ogun state nigeria. J Educ Pract. 2016; 7: 28-30.
- 2. Chandel K S, et al. Scientific attitude among senior secondary school students. Int J Appl Res. 2016; 2: 572-577.
- 3. Bai C A, et al. A study of scientific attitude of secondary school students in west tripura district. Int J Soc Sci. 2017:5: 608-618.
- 4. Balaji B, et al. Scholarly research journal for interdisciplinary studies. SJIF. 2016; 37: 1-2.
- 5. Rani B, et al. Science attitude analysis among secondary school students of yamunanagar region in relation to their achievement in general science, IJRAR. 2018; 5: 3-5.
- 6. Chhingpuii D. A study of science achievement, attitude toward science and problem solving ability among the secondary school students in aizawl. Int J Humanit Soc Sci. 2019; 4: 28-33.
- 7. Deboer, et al. Student attitudes and aspirations towards science. JO Second Int Handbook of Sci Edu1998;24:1585-1583.
- 8. Jampannanavar G C, et al. A study of the relationship between scientific attitude and academic achievement in science among secondary school students. Int Multidisciplinary Peer Rev Ref J. 2018; 7: 5-9.

- 9. Ahuja A, et al. A study carried on study of scientific attitude about science achievement scores among secondary school students' educational quest. Int J Education App SocSci. 2017; 8: 9-16.
- 10. Umashankar J, et al. Scientific attitude and academic achievement of secondary school students in coimbatore district. Shanlax Int J Educ. 2014; 2: 56-47.