

PHYSICO-CHEMICAL ANALYSIS OF GROUND WATER NEAR A THERMAL POWER PLANT

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Abstract: In the present study an attempt has been made to estimate the amount of heavy metals and pollutants getting accumulated in the ground water near the ash pond of thermal power plant near Talcher area, Angul. The study has been carried out keeping three season i.e. pre-monsoon, monsoon and post-monsoon. The chemical characterization of water samples have been carried out and are classified according to WHO guideline to estimate the impact of heavy metals and pollutants on ground water quality.

Keywords: Ash pond, thermal power plant, ground water, WHO guideline, heavy metals

I. INTRODUCTION

Water being the vital element of life, maintenance of a good water quality is highly essential for human wellbeing. Ground water has several applications in the field of cooking, domestic uses, industrial applications, agricultural utility and many more. Due to its purity and rich nutrient and mineral contain, ground water is preferred as a major source in the form of purist form of water (1-4). With the rapid modernization and huge growth in population, the demand of water is increasing day by day which in turn leading to disposal of this water to different major water sources such as sea, river etc. which in turn is contaminated the ground water sources.

Need of electricity for domestic as well as industrial use is enhancing day by day. To face this demand, generation of a huge quantity of electricity is found to be essential. In India the major sources for electricity generation through thermal power plant. Thermal Power Plant for electricity generation utilizes coal as the main source. In India availability of verities of coal has been made it possible for the existence of the thermal power plant.

In the present study the area taken into consideration is Talcher, Angul which is 70.78 meter above the main sea level. It is located between the latitude of 20^o29' and longitude of 34^o17'. The quantity of coal burn ash generated in this area is more than 172 Tons per year. The average rain fall is about 1123mm during monsoon. The areas under study are taken near by the thermal power plant by NTPC (TTPS) which is of 470 MW instillation capacities (5-8).

In the present study samples have been collected over a period of one year i.e. 2012 during pre-monsoon (February-March), monsoon (June-July) and post-monsoon (November-December) seasons from the area nearby to the ash pond of thermal power plant. The ground water quality is being collected from the tube wells and wells available there by used for drinking as well as industrial and agricultural applications.

II. MATERIALS AND METHOD

Water samples have been collected from different tube wells and other wells present near the ash pond area. The water level shows a drastically change from March to September i.e. adequate water get accumulated in the post-monsoon session due to heavy rain. The water samples were collected in stencilled bottles (polyethylene bottles). The test for pH, turbidity, water temperature and alkalinity has been carried out using multi functional water. The concentration of sodium and potassium are determined using Systronic Flame Photometer. The presence of temporary and permanent hardness is measured by EDTA method. Similarly the percentage of bio-carbonate is calculated by Bromo-cresol method whereas sulphate is done by gravimetric analysis and chloride by Mohr's method.

III. RESULT AND DISCUSSIONS

The entire data obtained is being tabulated in Table-1, 2 and 3 as mentioned herewith. The analysis result for pre-monsoon, monsoon and post-monsoon season are represented in Table-1,2and 3 respectively, show that the water sample is always found to be alkaline in nature varies within a range of 7.3 to 9.19.

TABLE I (HYDRO CHEMICAL COMPOSITION OF THE GROUND WATER SAMPLES DURING PRE-MONSOON-2012)

Village Name	pH	Conductivity	TDS	Alkalinity	Hardness	Cl	SO ₄ ⁻²	NO ₃ ⁻	Na	Ca	PO ₄ ⁻	HCO ₃ ⁻	Mg	K	Fe	F
Turanga	8.7	841	698	296	558	96	198	0.92	7.9	186	0.32	286	99	0.69	12737	186
Tulasipal	8.13	792	653	310	592	82	186	0.83	7.5	192	0.58	294	91	0.62	13132	193
Dera	8.4	748	614	284	514	78	152	0.86	7	178	0.23	210	86	0.58	11907	175
Gopinathpur	8.6	880	632	296	675	73	177	0.91	7.2	310	0.44	352	107	0.71	12342	201
Saradhapur	8.23	932	708	342	693	101	440	1.2	8.3	232	0.69	387	112	0.79	14132	222
Nuahata	8.52	772	668	288	560	85	382	0.87	8.1	184	0.47	358	102	0.68	13962	210
Kulad	8.02	801	672	318	610	92	410	0.99	7.8	305	0.62	312	89	0.72	11999	184

TABLE II (HYDRO CHEMICAL COMPOSITION OF THE GROUND WATER SAMPLES DURING MONSOON-2012)

Village Name	pH	Conductivity	TDS	Alkalinity	Hardness	Cl	SO ₄ ⁻²	NO ₃ ⁻	Na	Ca	PO ₄ ⁻	HCO ₃ ⁻	Mg	K	Fe	F
Turanga	8.86	1082	525	208	408	88	312	0.78	9.2	189	0.48	234	85.6	1.45	9723	114
Tulasipal	8.92	1153	610	212	413	122	286	0.65	8.7	197	0.39	218	67.2	1.32	10771	94
Dera	9.01	1282	498	231	383	109	220	0.68	8.6	202	0.36	221	74.1	1.61	11213	73
Gopinathpur	8.43	936	502	225	485	100	165	0.72	8.1	213	0.31	198	85.6	1.08	9478	122
Saradhapur	8.67	1126	472	248	306	112	102	0.53	8.3	136	0.42	187	91.3	1.14	8221	143
Nuahata	9.19	1198	623	236	345	92	171	0.41	9.0	142	0.38	191	72.4	1.23	10515	87
Kulad	8.99	1120	606	201	408	98	136	0.32	8.9	116	0.29	144	78.8	0.91	9899	82

MONSOON-2012)

TABLE III (HYDRO CHEMICAL COMPOSITION OF THE GROUND WATER SAMPLES DURING POST-MONSOON-2012)

<i>Village Name</i>	<i>pH</i>	<i>Conductivity</i>	<i>TDS</i>	<i>Alkalinity</i>	<i>Hardness</i>	<i>Cl</i>	<i>SO₄⁻²</i>	<i>NO₃⁻</i>	<i>Na</i>	<i>Ca</i>	<i>PO₄⁻</i>	<i>HCO₃⁻</i>	<i>Mg</i>	<i>K</i>	<i>Fe</i>	<i>F</i>
Turanga	7.3	672	801	348	701	132	398	0.96	9.01	411	0.77	413	196	0.61	14986	152
Tulasipal	7.5	701	842	355	695	96	286	0.82	9.78	312	0.81	299	172	0.73	15051	146
Dera	7.8	682	792	302	618	83	206	0.76	9.16	276	0.61	284	108	0.67	13962	138
Gopinathpur	7.6	768	873	364	644	128	213	0.71	10.40	377	0.69	315	153	0.83	14111	301
Saradhapur	8.5	878	898	372	732	143	414	1.5	10.42	423	0.89	428	200	0.91	16106	327
Nuahata	7.9	831	796	341	656	91	277	1.02	9.66	288	0.73	396	128	0.76	14781	271
Kulad	8.1	802	834	313	679	102	356	0.88	9.82	301	0.65	348	111	0.88	15051	233

The conductivity varies between 672-1282 moh/cm. the total dissolve solid is found to be in the range of 672-1282 mg/L. The TDS is found to be higher in post-monsoon which is attributed to be the accumulation of fly ash contaminated waste water into the ground water sources. The variation in alkalinity is between 201-372 mg/ L is slightly higher than the permissible limit whereas the nitrate contain is found to be between 0.78-8.3mg/L which is permissible as per the Who guideline. The variation in hardness, calcium, magnesium and sulphate are found to be higher than the permissible limit which may be attributed to the contamination of fly ash loaded surface water accumulated in ground water. The ranges are 306-732 mg/L in hardness 116-377mg/ L in calcium, 67.2-200mg/L in magnesium and in sulphate the range is varies from 102-440 mg/L.

The TDS gradually increases from pre-monsoon through monsoon to post-monsoon level which is attributed to the fact that with the rise growth in the plant, the variation in TDS has an adverse effect on the soil characteristics which in turns leads to the formation of insoluble complex compound that decreases the nutrients intake of the plant.

The chemical analysis data for other parameters such as sodium, potassium and iron are found to be in the range of 7-10.42 mg/L, 0.58-1.45 mg/L and 8221-15051 mg/L in iron respectively which may be due to the leaching of these nutrients more into the waste sources due to the increase in the pH level of the water. In case of Fluoride, it varies from 73-327 mg/L.

IV. CONCLUSION

The above study concluded that the ground water sources near by the ash pond are getting highly contaminated due to the leaching of different metals and toxic elements into the ground water sources. Keeping in view the health of plants, animals and human beings adequate measures are to be taken to get rid of ground water pollution.

REFERENCES

- [1] ISI Specification for Drinking Water, 1983, Indian Standard Institution, New Delhi.
- [2] Raghunath, H. M, Ground Water, Willy Eastern Limited Publication, 1990.
- [3] WHO guidelines for drinking water quality, vol-1, recommendation World Health Organization, Jeniva, 1980.
- [4] Piper, A.M.A, "Graphical Procedure in the Geochemical Interpretation of Water Analysis, U.S geological survey of ground water", 1953.
- [5] Stiff, H.A, "The Interpretation of Chemical Water Analysis by means of patterns", Journal of Petroleum Technology, Vol-3 (10), 1951.
- [6] Jain, C. K and Sharma, M. K, "Regression Analysis of ground water quality of Sagar district, Madhya Pradesh", Indian Journal of Environmental Health, vol-42, issue-4, 2000.
- [7] Tiwari, T. N and Ali, M. "Correlation among water quality parameters of Industrial waters", Indian Journal of Environmental Protection, vol-7, issue-14, 1988.
- [8] Kannan, N. and Rajasekharam, "Correlation of water quality parameters of painting industry effluents in Sivasakti, South India", Indian Journal of Environmental Health, vol-33 (3), 1991.