Research and Reviews: Journal of Chemistry

Flash Chromatography and Its Different Dissolvable Frameworks: A Review Dinesh Kumar G*

Department of pharmaceutical analysis and quality assurance, CMR College of pharmacy, India

Review Article

Received date: 26-06-2016 Accepted date: 20-07-2016 Published date : 26-07-2016

*For Correspondence: Dinesh Kumar G, Department of pharmaceutical analysis and quality assurance, CMR College of pharmacy, Medchal, Telangana, India, Tel: +919963330488; Email: <u>dineshkumar.2431@gmail.com</u>

Keywords: Silica gel, Flash chromatography, TLC, HPLC.

ABSTRACT

Flash chromatography is rapid form of preparative chromatography-preparative column liquid chromatography based upon an air pressure driven hybrid of medium and short column chromatography optimized for rapid separation of organic compounds. As technology has evolved available guidelines for normal-phase flash chromatography have become less relevant. Years of experience performing chromatography with disposable columns have been condensed into simple guidelines useful for translating TLC results into either isocraticgradient-flash or chromatography. The described studies should provide researchers with a means of selecting adequate columns and guidelines to reduce the waste of solvents, silica, time, and money. Modern flash chromatography systems are sold as prepacked plastic cartridges and the solvent is pumped through the cartridge. These systems may also be linked with detectors and fraction collectors providing automation. The introduction of gradient pumps has resulted in quicker separations and less solvent usage.

INTRODUCTION

Silica gel streak chromatography has gotten to be universal inside natural science and since its formal presentation in 1978 ^[1]. Every chromatographic technique except for TLC uses sections for the partition procedure. Section chromatography has discovered its place in numerous research centers for preparative purposes and for response control in natural amalgamations. The significance of segment chromatography is mostly because of taking after elements are given underneath i. Basic pressing strategy, ii. Low working weight, iii. Low cost for instrumentation ^[2].

Flash chromatography is fundamentally a pneumatic force driven half and half of medium weight and shorter

segment chromatography which has been enhanced for especially quick partition. Streak chromatography is a strategy used to separate blends of atoms into their individual constituents, as often as possible utilized as a part of the medication disclosure process. Flash chromatography varies from the customary procedure in two ways: to start with, somewhat littler silica gel particles 250-400 cross section are utilized, and second, because of limited stream of dissolvable brought about by the little gel particles, pressurized gas ca. 10-15 psi is utilized to drive the dissolvable through the section of stationary stage ^[3]. The net result is a fast "over in a glimmer" and high determination chromatography. A few makers have created computerized streak chromatography frameworks.

It characterized into two sorts 2:

1. LPLC - Low weight fluid chromatography LPLC framework which work around 50-75 psi

2. MPLC –Medium weight fluid chromatography MPLC frameworks which work above 150 psi. Mechanized blaze chromatography frameworks incorporate segments typically found on more costly HPLC frameworks, for example, a slope pump, test infusion ports, an UV indicator and a part gatherer to gather the eluent. Ordinarily these mechanized frameworks separate examples from a couple of milligrams up to a mechanical kg scale and offer much less expensive and speedier answer for doing different infusions on prep-HPLC frameworks. The product controlling a mechanized framework organize the segments, permit a client to just gather the groups that contain their objective compound accepting they are distinguishable on the framework's finder and help the client to discover the subsequent cleansed material inside the division gatherer. The product likewise spares the resulting chromatograph from the procedure for authentic and/or later review purposes.

Principle

The guideline is that the eluent is, under gas weight regularly nitrogen or compacted air quickly pushed through a short glass segment with substantial internal breadth. The glass section is stuffed with an adsorbent of characterized molecule size. The most utilized stationary stage is silica gel 40-63 µm, yet clearly pressing with other molecule sizes can be utilized also. Particles littler than 25 µm ought to be utilized with low consistency versatile stages on the grounds that generally the stream rate would be low. Typically gel beds are around 15 cm high with working weights of 1.5-2.0 bars. Initially just unmodified silica was utilized as the stationary stage, so that lone typical stage chromatography was conceivable. Meanwhile, be that as it may, and parallel to HPLC, switched stage materials are utilized all the more often as a part of glimmer chromatography.

Theory

Chromatography misuses the distinctions in apportioning conduct between a versatile stage and a stationary stage to isolate the segments in a blend. Mixes of the blend interface with the stationary stage in view of charge, relative solvency or adsorption. The maintenance is a measure of the velocity at which a substance moves in a chromatographic framework. In a ceaseless advancement framework like HPLC or GC where the mixes are eluted with the eluents, the maintenance is generally measured as the maintenance time rt, the time between the infusion and discovery. In un-interfered with improvement framework like TLC, the maintenance is measured as the maintenance element Rf, the run length of the compound partitioned by the run length of the eluent front. Rf=Distance went by the dissolvable front ^[4,5].

Different parts of Flash Chromatographic System

- The essential for effective partitions is the decision of the best possible adsorbent. The most essential stationary stage in segment chromatography is silica.
- Silica gel SiO₂ and alumina Al₂O₃ are two adsorbents ordinarily utilized by the natural scientist for section chromatography. These adsorbents are sold in various cross section sizes, as showed by a number on the jug mark: "silica gel 60" or "silica gel 230-400" is two or three illustrations.
- This number eludes to the lattice of the sifter used to estimate the silica, particularly, the quantity of openings
 in the cross section or strainer through which the unrefined silica molecule blend is passed in the assembling
 procedure. Adsorbent molecule size influences how the dissolvable courses through the section. Littler
 particles higher cross section qualities are utilized for glimmer chromatography; bigger particles lower network
 qualities are utilized for gravity chromatography.
- For instance, 70-230 silica gels are utilized for gravity segments and 230-400 lattices for glimmer segments. The measure of silica gel relies on upon the Rf distinction of the mixes to be isolated, and on the measure of test. For n grams of test, you ought to utilize 30 to 100 n grams of silica gel.
- For less demanding divisions, proportions more like 30: 1 are powerful, for troublesome partitions, more silica gel is regularly required. In any case, by utilizing more silica gel, the time span required for the chromatography is expanded. The thickness of powdered silica gel is around 0.75 g for every ml.
- These are a few adsorbents which are essentially utilized as a part of blaze chromatography [6].
- Silica: Slightly acidic medium. Best for normal mixes, great partition is accomplished.
- Florisil: Mild, nonpartisan medium. 200 cross section can be compelling for simple divisions. Under 200 work best for cleaning by filtration. A few mixes stick on florisil, test first.
- Alumina: Basic or impartial medium. Can be compelling for simple partitions, and sanitization of amines.
- Reverse stage silica: The most polar mixes elute quickest, the most nonpolar slowest. Dissolvable Systems.
- Streak segment chromatography is generally completed with a blend of two solvents, with a polar and a nonpolar segment Figure 1^[7].

One-Part Dissolvable Frameworks

- 1. Hydrocarbons: pentane, petroleum ether, hexanes
- 2. Ether and dichloromethane fundamentally the same as extremity
- 3. Ethyl acetic acid derivation

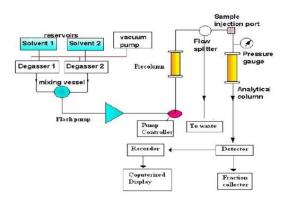


Figure 1. Instrumentation of Flash Chromatography.

Two-part dissolvable frameworks

- i. Ether/Petroleum Ether, Ether/Hexane, and Ether/Pentane: Choice of hydrocarbon part relies on accessibility and prerequisites for bubbling reach. Pentane is costly and low-bubbling, petroleum ether can be low-bubbling, and hexane is promptly accessible.
- ii. Ethyl Acetate/Hexane: The standard, useful for conventional mixes and best for troublesome divisions.
- iii. Methanol/Dichloromethane: For polar mixes.
- iv. 10 % Ammonia in Methanol Solution/Dichloromethane: Sometimes moves persistent amines off the benchmark.
- v. For fundamental i.e. nitrogen containing mixes, it is at times valuable or important to include a little measure of triethylamine or pyridine to the dissolvable blend around 0.1%.
- vi. For acidic aggravates, a little measure of acidic corrosive is at times valuable. For this situation, be extremely watchful in concentrating the dissolvable as follow measures of acids can be exceptionally perilous when they are concentrated with an item. In these cases, the acidic corrosive can regularly be securely rotavaped away by including parts of toluene and concentrating to a couple mL volumes and rehashing this few times. As acidic corrosive bubbles at a lower BP than toluene, this will expel the corrosive without uncovering the flawless compound to it. The properties of regularly utilized glimmer solvents. The compound of interest ought to have a TLC Rf of ≈ 0.15 to 0.20 in the dissolvable framework you pick. Double two segment dissolvable frameworks with one dissolvable are having a higher extremity than the other are typically best since they take into account simple modification of the normal extremity of the eluent. The proportion of solvents decides the extremity of the dissolvable framework, and thus the rates of elution of the mixes to be isolated. Higher extremity of dissolvable expands rate of elution for all mixes. In the event that your Rf is a ≈ 0.2 , you will require a volume of dissolvable $\approx 5X$ the volume of the dry silica gel so as to run your segment Table 1.

| Solvent | Density g/ml | Elution Strength | Solvent Group | Boiling Point °C | UV Cut- off nm | TLV ppm |
|--------------------------|-----------------|---------------------|------------------|---------------------|-------------------|---------|
| n-Hexane | 0.66 | 0.01 | 1 | 69 | 195 | 100 |
| 2 2 4-Trimethylpentane | 0.69 | 0.02 | 1 | 99 | 210 | 300 |
| Cyclohexane | 0.77 | 0.03 | 1 | 81 | 200 | 100 |
| 112- Trichloromethane | 1.48 | 0.31 | 8 | 61 | 245 | 50 |
| Toluene | 0.87 | 0.22 | 7 | 110 | 285 | 100 |
| Dichloromethane | 1.33 | 0.3 | 5 | 40 | 232 | 100 |
| Ethyl Acetate | 0.9 | 0.45 | 6 | 77 | 256 | 400 |
| Methyl-t-butyl ether | 0.74 | 0.48 | 2 | 55 | 210 | 40 |
| Acetone | 0.79 | 0.53 | 6 | 56 | 330 | 750 |
| Tetrahydrofuran | 0.89 | 0.35 | 4 | 66 | 212 | 200 |
| Acetonitrile | 0.78 | 0.5 | 6 | 82 | 190 | 40 |
| Isopropanol | 0.79 | 0.6 | 3 | 82 | 205 | 400 |
| Ethanol | 0.79 | 0.88 | 3 | 78 | 210 | 1000 |

| Table 1. | Different | solvents of | of parameters. |
|----------|-----------|-------------|----------------|
|----------|-----------|-------------|----------------|

| Methanol | 0.79 | 0.7 | 3 | 65 | 205 | 200 |
|----------|------|-------|---|-----|-----|-----|
| Water | 1 | 0.073 | 8 | 100 | 180 | - |

Section Selection

Select a section that is 10, 20, 40 mm ID based upon preparative necessities. In fact, Professor Still et al offered this choice Table 2: Single Step Flash Columns protected speak to a creative stride forward in chromatography. Streak Chromatography is a speedy and cheap method for the cleansing of natural mixes. Thomson streak segments arrive in a wide assortment of sizes going from 4g to 300g silica-based for simple adaptability of manufactured responses. Thomson additionally offers other pressing material like Amine and C18 streak segments which empower the end-client to use these blaze sections for a wide scope of responses.

| Column | Volume of | Sample Load mg | | Fraction |
|------------|-----------|----------------|----------|----------|
| Diametermm | eluent ml | Rf < 0.2 | Rf > 0.1 | Size ml |
| 10 | 100 | 100 | 40 | 5 |
| 20 | 200 | 400 | 160 | 10 |
| 30 | 400 | 900 | 360 | 20 |
| 40 | 600 | 1600 | 600 | 30 |
| 50 | 1000 | 2500 | 1000 | 50 |

Table 2. Typical volume of eluent required for packing and elution.

Dissolvable Selectivity

Dissolvable selectivity is characterized as the dissolvable to specifically influence the maintenance of one compound in the blend with respect to the others, consequently influencing Δ Rf and CV. Solvent selectivity ought to be acclimated to give Δ Rf > 0.20. Diverse dissolvable mixes to get fancied TLC detachment for the most part uncovers fitting conditions for successful blaze chromatography partition. Diverse dissolvable blends can even invert the elution request of a portion of the segments in the example. Segment volume contrast Δ CV predicts section limit or the measure of material that can be viably isolated in a solitary segment stacking. The more noteworthy the Δ CV, the more noteworthy the successful limit of the section.

METHODOLOGY

Pressing the Column



Figure 2. Pressing the cloumn.

A chromatography segment is stopped with a little bit of cotton fleece, sufficiently only to fill to stopcock opening. Sand, around 2 cm is included so that the distance across of the sand is roughly the same as the section Figure 2. Silica gel is included dry. More often than not, it is ideal if the silica is not very long, around 6 to 10 inches is best much of the time. Join the house vacuum to the base of the section by means of the stopcock. Open the vacuum and the stopcock; this will packs the silica gel and hold it tight for the following strides. Add sand to the highest point of the section, around 1-2 cm is sufficient. With the vacuum still connected, pour the dissolvable premixed, i.e. 4:1 hexanes/ethyl acetic acid derivation. Permit the dissolvable to stream however the segment until it is practically eluting. Now, shut the stopcock and expel the vacuum line. Ensure enough dissolvable is in the segment for 5-6 segment volumes worth to stream however, to guarantee complete pressing. Presently elute the greater part of the dissolvable with pneumatic force, taking consideration not the let the section run dry. Stop with the dissolvable level parallel with the sand. An all-around stuffed segment ought not to have any splits or fixes. The dissolvable escaping from the stopcock ought not be warm or hot ^[8].

Stacking the Column

Set up an answer of your response or compound blend in the insignificant measure of methylene chloride conceivable. Utilizing a pipette, add this painstakingly to the highest point of the silica, washing the carafe 3-4 times with methylene chloride or the chromatography dissolvable. After every option, permit the dissolvable level to slide into the extremely top of the silica gel beneath the sand.Carefully included 2-3 pipettes of chromatography dissolvable and push this into the section rehash 3-4x.Now, deliberately fill the rest of the segment space with the chromatography dissolvable and elute utilizing packed air. A stream rate of around 2 inches/moment is perfect. This is measured by how quick the dissolvable section plunges in the straight piece of segment, over the silica gel. It is most advantageous to gauge and alter the stream rate before including the compound. In situations where a response blend or compound is not dissolvable in a reasonable dissolvable for stacking, it can be ingested onto silica gel. This is finished by dissolving the compound in CH₃₂CO, including silica gel and painstakingly focusing the silica gel to dryness watchful: it knocksl. The dry silica is then added to the highest point of the stuffed silica section. For this situation, sand ought not to be added to the section until after the silica-compound blend is included. This technique is suggested just if all else fails as partitions are frequently mediocre compared to arrangement stacking ^[3].

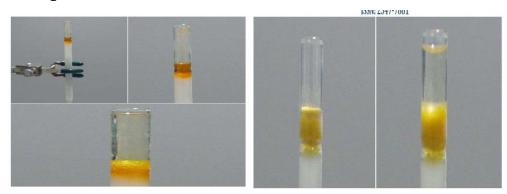


Figure 3. Stacking the Column.

Running the segment

Segment portions are gathered in test tubes, of a size suitable for the kind of segment and extremity. Utilize the 13 mm test tubes for little scale i.e. 5-50 mg and bigger test tubes for greater segments. Allude to the rules in Still's paper for picking portion sizes. Start gathering the division instantly in the wake of including your compound;

it doesn't take long for extremely non-polar mixes to elute from the column. Once you have stacked a section, it is best not to stop it for any period of time. This is because of moderate dissemination of the mixes on the silica gel bringing about poor division and lessened yields. To discover your item, detect every part or so on a TLC plate and check which portions contain mixes. Parts containing the same mixes are joined, the test tubes washed with methylene chloride or most likely better for nature, refined ethyl acetic acid derivation, and the dissolvable concentrated under lessened weight. Try not to give a section a chance to run dry or elute the dissolvable until after you are certain the majority of the mixes have eluted ^[9-18].

After the section tidying up

After you have completed, elute the greater part of the dissolvable from the section utilizing compacted air. Streaming air through the segment for ~2 hours will give dry, free streaming silica gel. Spill out the substance of the segment into the silica waste holder. By and large, washing the segment with water and $CH_{32}CO$ is adequate. On the off chance that vital, a little measure of fluid cleanser can be utilized. Attempt to abstain from scratching the segments with rough brushes or cleansers ^[19-30].

Utilization of Flash Chromatography

Normal mixes are increasingly assessed as other options to established medications and in this way the requirements for the partition of such complex blends are additionally developing. Streak chromatography is an extremely significant system in the field of regular mixes research since it gives a quick and temperate approach to isolate the fundamental segments of complex plant separates.

Streak chromatography is the regularly utilized filtration apparatus after natural blend. The "Short Notes accompanying" outline the adaptability of the Sepacore framework for the enhancement of the blaze purging of different combination response blends. Confinement of 4-Methoxyacetophenone from a rough response blend, lsocratic elution, utilizing an extra polar dissolvable for test stacking, lsolation of Benzoin from an unrefined response blend, Cleaning up of $\alpha\alpha\alpha$ -Methyl styrene by blaze chromatography, lsolation of 3-Nitro-4-ethoxybenzaldehyde from an amalgamation blend, lsolation of Benzylideneacetophenone from an unrefined response blend [31-60].

CONCLUSION

Streak Chromatography is a basic, quick, financially savvy Preparative Liquid Chromatography approach. Partitions are based upon customarily gotten TLC results which are essentially extrapolated to preparative scale. Streak chromatography is exceptionally helpful method for rapidly isolating expanding amounts of tests. It is unsurprising and simple to scale here and there as required. Present day instrumentation is making it less demanding still to take full control over the detachment and the system keeps on growing rapidly ^[61-99].

REFERENCES

- 1. Roge AB, et al. Brief Review On: Flash Chromatography. Int J Pharma Sci Res. 2011; 28: 5-11.
- 2. www.chem.rochester.edu/how to flash.html
- 3. www.wapedia.in
- 4. www.sorbeadindia.com

- Cox GB and Snyder LR. Preparative high-performance liquid chromatography under isocratic conditions. II. The consequences of two adjacent bands having unequal column capacities. J Chromatogr 1989; 483: 95-110.
- 6. Chattopadhyay SK. Flash chromatography and low pressure chromatographic techniques for separation of phytomolecule. Central Institute of Medicinal and Aromatic Plants cimap, Lucknow.
- 7. Ahuja S and Scypinski S. Handbook of Modern Pharmaceutical Analysis. 122-123p.
- 8. McGuffin VL. Chromatography. (6th ed) Elsevier, Oxford, UK, 2004.
- 9. Reach Device US-made scientific equipment, The First Usable Detector for Flash Chromatography.
- 10. Still WC, et al. Flash chromatography. J Org Chem. 1978; 4314: 2923-2925.
- 11. Stout RW, et al. High-performance liquid chromatographic column efficiency as a function of particle composition and geometry and capacity factor. J Chromatogr. 1983; 282: 263-286.
- 12. Furniss BJ and Hannuford AS. Vogels Text book of Organic Chemistry. (5th ed) 217-219.
- 13. William CS and Hill DC. General methods for flash chromatography using disposable column. Mol Divers 2009; 132: 247-252.
- 14. www.biotage.com
- 15. www.pretech.nu/products
- 16. www.pretech.nu/product
- 17. Buchi Preparative Chromatography
- Piteni Al, et al. HILIC Chromatography-An Insight on the Retention Mechanism. J Chromatogr Sep Tech.
 2016; 7: 326
- 19. Wakamoto H and Miyamoto M. Development of a New Dermatophyte-Detection Device using Immunochromatography. J Med Diagn Meth. 2016; 5: 216.
- 20. Mahendra Kumar T, et al. Evaluation of the Isotopic Abundance Ratio in Biofield Energy Treated Resorcinol Using Gas Chromatography-Mass Spectrometry Technique. Pharm Anal Acta. 2016; 7: 481.
- 21. Michalski R. Ion Chromatography and Related Techniques 2016. J Chromatogr Sep Tech. 2016; 7: 325.
- 22. Arnoldi S, et al. Validation Study of Analysis of 1-Phenyl-2-Propanone in Illicit Methamphetamine Samples by Dynamic Headspace Gas Chromatography Mass Spectrometry. J Chromatogr Sep Tech. 2016; 7: 322.
- 23. Ivkovic B, et al. Chemometrical Evaluation of Metoprolol Tartarate Enantiomers Separation Applying Conventional Achiral Chromatography. J Anal Bioanal Tech. 2016; 7: 303.
- 24. Heidari A. Measurement the Amount of Vitamin D₂ Ergocalciferol, Vitamin D₃ Cholecalciferol and Absorbable Calcium Ca²⁺, Iron II Fe²⁺, Magnesium Mg²⁺, Phosphate PO₄- and Zinc Zn²⁺ in Apricot Using High-Performance Liquid Chromatography HPLC and Spectroscopic Techniques. J Biom Biostat. 2016; 7: 292.
- 25. Musirike MR, et al. Stability Indicating Reverse Phase Chromatographic Method for Estimation of Related Substances in Voriconazole Drug Substance by Ultra Performance Liquid Chromatography. Pharm Anal Acta. 2016; 7: 460.
- 26. Nimmanwudipong T, et al. Determination of Intramolecular ¹³C Isotope Distribution of Pyruvate by Headspace Solid Phase Microextraction-Gas Chromatography-Pyrolysis-Gas Chromatography-Combustion-Isotope Ratio Mass Spectrometry HS-SPMEGC-Py-GC-C-IRMS Method. J Anal Bioanal Tech. 2015; 7: 293.
- 27. Kuvshinova SA, et al. Selectivity, Thermodynamic and Anisotropic Properties of Substituted Liquid-Crystal Cyanoazoxybenzenes as Stationary Phases for Gas Chromatography. J Chromatogr Sep Tech. 2016; 7:

314.

- 28. Guo WR, et al. Simultaneous Detection Method for Mycotoxins and their Metabolites in Animal Urine by Using Impurity Adsorption Purification followed by Liquid Chromatography-Tandem Mass Detection. J Chromatogr Sep Tech. 2015; 6: 308.
- Linnerz K, et al. Liquid Chromatography-Tandem Mass Spectrometry Method for the Quantification of Fentanyl and its Major Metabolite Norfentanyl in Critically III Neonates. J Chromatograph Separat Techniq. 2015; S6: 004.
- 30. Gineys M, et al. Simultaneous Determination of Pharmaceutical and Pesticides Compounds by Reversed Phase High Pressure Liquid Chromatography. J Chromatogr Sep Tech. 2015; 6: 299.
- 31. Justiz-Vaillant AA, et al. Purification of the Mule Equus mulus IgG by Protein A-Affinity Chromatography. J Chromatogr Sep Tech. 2015; 6: 298.
- Gritti F. Retention Mechanism in Hydrophilic Interaction Liquid Chromatography New sights Revealed From the Combination of Chromatographic and Molecular Dynamics Data. J Chromatogr Sep Tech. 2015; 6: 309.
- 33. Chauhan MK and Bhatt N. A Simple and Modified Method Development of Vancomycin Using High Performance Liquid Chromatography. J Chromatogr Sep Tech. 2015; 6: 296.
- Amini A. Identification of E-Caprolactam and Melamine in Polyvinyl-Pyrrolidone Powder by Double Injection
 Micellar Elektrokinetic Chromatography.Pharm Anal Acta. 2015; 6: 442.
- Nakano T and Ozimek L. Selective Removal of Phenylalanine Impurities from Commercial κ-Casein Glycomacropeptide by Anion Exchange Chromatography. J Food Process Technol. 2015; 7: 537.
- 36. Wang JM, et al. Analysis of Fructose 1,6-Diphosphate in Fermentation Broth Using Ion Chromatography. Biochem Anal Biochem. 2015; 4: 209.
- 37. Salvatierra-Stamp VC, et al. Supercritical-Fluid Chromatography with Diode-Array Detection for Emerging Contaminants Determination in Water Samples. Method Validation and Estimation of the Uncertainty. J Chromatogr Sep Tech. 2015; 6: 291.
- Chen Z, et al. Utilization of a Matrix Effect to Enhance the Sensitivity of Residual Solvents in Static Headspace Gas Chromatography. J Chromatogr Sep Tech. 2015; 6: 289.
- Goswami J. Different Separation or Experimental Techniques for Clinical Chromatography: Small Review. J Chromatogr Sep Tech. 2015; 6: 297.
- 40. Willmann L, et al. Comprehensive Two-Dimensional Liquid Chromatography in Metabolome Analysis. J Chromatogr Sep Tech. 2015; 6: 288.
- 41. Bokhart M, et al. Determination of Organochlorine Pesticides in Wildlife Liver and Serum Using Gas Chromatography Tandem Quadrupole Mass Spectrometry. J Chromatogr Sep Tech. 2015; 6: 286.
- 42. Gupta A, et al. Determination of Quercetin a Biomarker in Hepatoprotective Polyherbal Formulation through High Performance Thin Layer Chromatography. J Chromatogr Sep Tech. 2015; 6: 285.
- 43. Albert K, et al. Investigating Insect Adhesion Secretions by Gas Chromatography-Mass Spectrometry. J Chromatograph Separat Techniq. 2015; S6: 001.
- Bargańska Ż, et al. Development of a Gas Chromatography-Tandem Mass Spectrometry Procedure for Determination of Pesticide Residues in Honey and Honeybee Samples. J Chromatograph Separat Techniq. 2015; S6: 002.

- 45. Trivedi MK, et al. Investigation of Isotopic Abundance Ratio of Biofield Treated Phenol Derivatives Using Gas Chromatography-Mass Spectrometry. J Chromatograph Separat Techniq. 2015; S6: 003.
- 46. Trivedi MK, et al. Isotopic Abundance Analysis of Biofield Treated Benzene, Toluene and p-Xylene Using Gas Chromatography-Mass Spectrometry GC-MS. Mass Spectrom Open Access. 2015; 1: 102.
- 47. Chakravarti B and Chakravarti DN. Liquid Chromatography-Tandem Mass Spectrometry-Application for Clinical Chemistry Laboratory. J Mol BiomarkDiagn. 2015; 6: 244.
- EL-Maali NABO and Wahman AY. Gas Chromatography-Mass Spectrometric Method for Simultaneous Separation and Determination of Several Pops with Health Hazards Effects. Mod Chem appl. 2015; 3: 167.
- 49. Stephen S, et al. Tracking Interfacial Adsorption/Desorption Phenomena in Polypropylene/Biofuel Media using Trace Cr³⁺/Cr⁶⁺ and As³⁺/As⁵⁺-A Study by Liquid Chromatography-plasma Mass Spectrometry. J Pet Environ Biotechnol. 2015; 6: 239.
- 50. Belico de VA, et al. Gel Filtration Chromatography Technique as Tool of Simple Study Seminal Plasma Proteins in Domestic Animals. J Chromatogr Sep Tech. 2015; 6: 281.
- Bernardi T, et al. Separation and Quantitative Determination of Carbohydrates in Microbial Submerged Cultures Using Different Planar Chromatography Techniques HPTLC, AMD, OPLC. J Anal Bioanal Tech. 2015; 6: 250.
- 52. Sanaki T, et al. Improvements in the High-Performance Liquid Chromatography and Extraction Conditions for the Analysis of Oxidized Fatty Acids Using a Mixed-Mode Spin Column. Mod Chem appl. 2015; 3: 161.
- 53. Delhiraj N and Anbazhagan S. A Simple, Isocratic and Ultra-Fast Liquid Chromatography/Mass Spectrometry Method for the Estimation of Barnidipine in Human Plasma. Pharm Anal Acta. 2015; 6: 400.
- 54. Amagai T, et al. Determination of Nicotine Exposure Using Passive Sampler and High Performance Liquid Chromatography. Pharm Anal Acta. 2015; 6: 399.
- 55. Guzel M, et al. Estimation of Octanol-Water Partition Coefficient Using Cationic Gemini Surfactants by Micellar Electrokinetic Chromatography. J Chromatogr Sep Tech. 2015; 6: 275.
- 56. Santini DA, et al. Development of a High Performance Liquid Chromatography Method for the Determination of Tedizolid in Human Plasma, Human Serum, Saline and Mouse Plasma. J Chromatogr Sep Tech. 2015; 6: 270.
- 57. Hee KH, et al. Development and Validation of Liquid Chromatography-Tandem Mass Spectrometry Method for the Quantification of Ertapenem in Human Serum. Pharm Anal Acta. 2015; 6: 358.
- 58. Karl WKS, et al. A Pharmacokinetic Analyses of Ferulic Acid in Rat Plasma by Liquid Chromatography– Tandem Mass Spectrometry: A Synergistic Action of an Ancient Herbal Decoction Fo Shou San. Pharm Anal Acta. 2015; 6: 361.
- 59. Ulmer CZ, et al. Liquid Chromatography-Mass Spectrometry Metabolic and Lipidomic Sample Preparation Workflow for Suspension-Cultured Mammalian Cells using Jurkat T lymphocyte Cells. J Proteomics Bioinform. 2015; 8: 126-132.
- 60. AL-Jammal MKH, et al. Development and Validation of Micro Emulsion High Performance Liquid Chromatography MELC Method for the Determination of Nifedipine in Pharmaceutical Preparation. Pharm Anal Acta. 2015; 6: 347.

- 61. Wang SW, et al. Simultaneous Quantitative Determination of Nine Bufadienolides in Traditional Chinese Medicinal Toad Skin from Different Regions of China by High-Performance Liquid Chromatography– Photodiode Array Detection. Pharm Anal Acta. 2015; 6: 345.
- 62. Wujian J, et al. A Simple Protein Precipitation-based Simultaneous Quantification of Lovastatin and Its Active Metabolite Lovastatin Acid in Human Plasma by Ultra-Performance Liquid Chromatography-Tandem Mass Spectrometry using Polarity Switching. J Chromatogr Sep Tech. 2015; 6: 268.
- 63. Maher HM. Stacking As Sample On-Line Pre-concentration Technique in Microemulsion Electrokinetic Chromatography. J Chromatogr Sep Tech. 2015; 6: e130.
- 64. Steiner WE and English WA. Emerging Trends in Gas Chromatography and Mass Spectrometry Instrumentation for Analytical & Bioanalytical Techniques. J Anal Bioanal Tech. 2015; 6: e118.
- 65. Black SM, et al. Quantitative Analysis of L-Abrine and Ricinine Spiked into Selected Food Matrices by Liquid Chromatography-Tandem Mass Spectrometry. J Chromatogr Sep Tech. 2015; 6: 265.
- 66. Dare M, et al. Method Validation for Stability Indicating Method of Related Substance in Active Pharmaceutical Ingredients Dabigatran Etexilate Mesylate by Reverse Phase Chromatography. J Chromatogr Sep Tech. 2015; 6: 263.
- 67. Silva MLS. Comprehensive Analysis of Phytopharmaceutical Formulations-An Emphasis on Two-Dimensional Liquid Chromatography. J Chromatogr Sep Tech. 2015; 6: 262.
- 68. P Ayare, et al. Flash Chromatography: Area & Applications. Pharma Tutor. 2014; 25: 89-103.
- 69. William C, et al. General methods for flash chromatography using disposable columns. Mol Divers. 2009; 132; 247-252.
- 70. Gregory R, et al. A Modern Apparatus for Performing Flash Chromatography: An Experiment for the Organic Laboratory. J Chem Educ. 2013; 903: 376-378.
- Sarah Millar. Tips and Tricks for the Lab: Column Troubleshooting and Alternatives. Chem views Magzine.
 2012.
- 72. Clark Still W, et al. Rapid chromatographic technique for preparative separations with moderate resolution. J Org Chem. 1978; 4314: 2923–2925.
- 73. Yan Y, et al. HPLC-DAD-Q-TOF-MS/MS analysis and HPLC quantitation of chemical constituents in traditional Chinese medicinal formula Ge-Gen Decoction. J Pharm Biomed Anal. 2013; 80: 192.
- 74. Zhou JL, et al. Herbal medicine analysis by liquid chromatography/time-of-flight mass spectrometry. J Chromatogr A. 2009; 1216: 7582-7594.
- 75. Qu CL, et al. Studies on fragmentation pathways of amino acids and their interactions with ginsenoside Rb3by spectrospray ionization mass spectrometry. Chem J Chinese U. 2008; 9: 1721-1726.
- Huang YF and Hu J. Simultaneous analysis of twenty free amino acids in tobacco using liquid chromatography-electrospray ionization/iontraps tandem mass spectrometry. Chin J Chromatogr. 2010; 6: 615-622.
- 77. Wang Y, et al. Fragmentation characteristics and utility of ammonium ions for peptide identification by MALDI TOF/TOF spectrometry. Chinese J Anal Chem. 2014; 7: 1010-1016.
- 78. Daniel D, et al. Determination of biogenic amines in beer and wine by capillary electrophoresis-tandem mass spectrometry. J Chromatogr A. 2015; 1416: 121-128.
- 79. Wu YL, et al. Simultaneous determination of sixteen amide fungicides in vegetables and fruits by dispersive solid phase extraction and liquid chromatography-tandem mass spectrometry. J Chromatogr B.

2015; 989: 11-20.

- 80. Zoran K, et al. Liquid chromatography tandem mass spectrometry method for characterization of monoaromatic nitro-compounds in atmospheric particulate matter. J Chromatogr A. 2012; 1268: 35-43.
- 81. Sun Y, et al. Qualitative and quantitative analysis of phenolics in Tetrastigma hemsleyanum and their antioxidant and anti-proliferative activities. J Agric Food Chem. 2013; 61: 10507-10515.
- 82. Fu Y, et al. Characterization and identification of baccharane glycosides in Impatientis Semen by rapidresolution liquid chromatography with electrospray ionization quadrupole time-of-flight tandem mass spectrometry. J Pharm Biomed Anal. 2012; 65: 64-71.
- 83. Chen XF, et al. Liquid chromatography coupled with time-of-flight and ion trap mass spectrometry for qualitative analysis of herbal medicines. J Pharmaceut Ana. 2011; 4: 235-245.
- 84. Alpert AJ. Hydrophilic-interaction chromatography for the separation of peptides, nucleic acids and other polar compounds. J Chromatogr. 1990; 499: 177-196.
- 85. Heaton J, et al. Comparison of reversed-phase and hydrophilic interaction liquid chromatography for the separation of ephedrines. J Chromatogr A. 2012; 1228: 329-337.
- 86. Buszewski B and Noga S. Hydrophilic interaction liquid chromatography HILIC--a powerful separation technique. Anal Bioanal Chem. 2012; 402: 231-247.
- 87. Tache F, et al. Greening pharmaceutical applications of liquid chromatography through using propylene carbonate-ethanol mixtures instead of acetonitrile as organic modifier in the mobile phases. J Pharm Biomed Anal. 2013; 75: 230-238.
- 88. Dos Santos Pereira A, et al. The acetonitrile shortage: is reversed HILIC with water an alternative for the analysis of highly polar ionizable solutes? J Sep Sci. 2009; 32: 2001-2007.
- Dai J and Carr PW. Effect of mobile phase anionic additives on selectivity, efficiency, and sample loading capacity of cationic drugs in reversed-phase liquid chromatography. J Chromatogr A. 2009; 1216: 6695-6705.
- 90. Bernal J, et al. Hydrophilicinteraction liquid chromatography in food analysis. J Chromatogr A. 2011; 1218: 7438-7452.
- 91. Hemstrom P and Irgum K. Hydrophilic interaction chromatography. J Sep Sci. 2006; 29: 1784-1821.
- 92. Jandera P. Stationary and mobile phases in hydrophilic interaction chromatography: a review. Anal Chim Acta. 2011; 692: 1-25.
- McCalley DV. Is hydrophilic interaction chromatography with silica columns a viable alternative to reversed-phase liquid chromatography for the analysis of ionisable compounds? J Chromatogr A. 2007; 1171: 46-55.
- 94. McCalley DV. Evaluation of the properties of a superficially porous silica stationary phase in hydrophilic interaction chromatography. J Chromatogr A. 2008; 1193: 85-91.
- 95. Hao Z, et al. Impact of column temperature and mobile phase components on selectivity of hydrophilici nteraction chromatography HILIC. J Sep Sci. 2008; 31: 1449-1464.
- 96. Buckenmaier SM, et al. Overloading study of bases using polymeric RP-HPLC columns as an aid to rationalization of overloading on silica-ODS phases. Anal Chem. 2002; 74: 4672-4681.
- 97. Po HN and Senozan NM. Henderson-Hasselbalch Equation: Its History and Limitations. J Chem Educ. 2001; 78: 1499-1503.
- 98. McCalley DV. Study of the selectivity, retention mechanisms and performance of alternative silica-based

stationary phases for separation of ionised solutes in hydrophilic interaction chromatography. J Chromatogr A. 2010; 1217: 3408-3417.

99. Kouskoura MG, et al. Elucidation of the retention mechanism on a reverse-phase cyanocolumn by modeling. J Sep Sci. 2014; 37: 1919-1929.